

Environmental Health

Carcinogenesis

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Neoplasia is new **growth or autonomous growth of tissue**. The resulting neoplastic lesion is referred to as a neoplasm. Both **benign and malignant neoplasms** can be induced by chemical carcinogens

The term “**tumor**” describes a lesion that may or may not be neoplastic, and is characterized by **swelling or an increase in size**

The term “**cancer**” describes the subset of **neoplasia** that represents **malignant neoplasms**. A carcinogen is an agent, chemical or physical, that causes or induces neoplasia

Carcinogens can be chemicals, viruses, hormones, radiation, or solid material

Malignant vs. Benign Neoplasms

- **Benign**

- Usually encapsulated
- Usually non-invasive
- Highly differentiated
- Rare mitoses
- Slow growth
- Little or no anaplasia
- No metastases

- **Malignant**

- Encapsulated
- Invasive
- Poorly differentiated
- Mitoses relatively common
- Rapid growth
- Anaplastic to varying degrees
- Metastases

Cancer is therefore the malignant **uncontrolled** proliferation of neoplastic cells.

Also a description of a multitude of different disease states (~200)

Cancer is the leading disease-related cause of years of life lost in the US.

- | • Causes of Death | • Years of Life Lost* |
|--------------------------|------------------------------|
| – All causes | – 11,761,000 |
| – Unintentional injuries | – 2,306,000 |
| – Cancer | – 1,803,000 |
| – Heart disease | – 1,563,000 |
| – Suicide, homicide | – 1,247,000 |
| – Congenital anomalies | – 584,000 |

* Estimated years of life lost before the age of 65

Features of Genotoxic and Nongenotoxic Carcinogens

Genotoxic carcinogens Mutagenic Can be complete carcinogens

Nongenotoxic carcinogens **Nonmutagenic**, reversible Tumorigenicity is dose responsive No direct DNA damage Species, strain, tissue specific

MULTISTAGE CARCINOGENESIS

Initiation

The first stage of the cancer process involves initiation, a process that is defined as a **stable, heritable change**. This stage is a **rapid, irreversible** process that results in a carcinogen-induced mutational event.

Initiating agents lead to genetic changes including mutations and deletions. **Chemical carcinogens that covalently bind to DNA and form adducts** that result in mutations are initiating agents.

Included among chemicals classified as initiating carcinogens are compounds such as **polycyclic hydrocarbons** and **nitrosamines**, **biological agents** such as viruses, and physical agents such as X-rays and UV light.

Most chemical carcinogens that function at the initiation stage of the cancer process are **indirect-acting genotoxic compounds that require metabolic activation** in the target cell to produce the DNA-damaging event.

The **ultimate** form of the carcinogen is then able to bind to nuclear DNA, resulting **in adduct formation**. The initiating event becomes “fixed” when the DNA adducts or other damage to DNA are **not correctly repaired**.

Promotion

Derived from either **endogenous or exogenous stimuli** of cell growth, the second stage of the carcinogenesis process involves the selective clonal expansion of initiated cells to produce a **preneoplastic lesion**.

Tumor promoters are not mutagenic and generally **are not able to** induce tumors by themselves; rather they act through several mechanisms involving **gene expression** changes that result in **sustained** cell **proliferation**, either through increases **in cell proliferation and/or the inhibition of apoptosis**.

Promotion is a **reversible** phenomenon where by upon removal of the promoting agent, the focal cells may return to **single initiated cell threshold**

Tumor promoters in general **show organ-specific effects** , e.g.,a tumor promoter of the liver, such as **phenobarbital** will not function as a tumor promoter in the skin or other tissues.

Progression

The final stage of the carcinogenesis process, progression, involves the conversion of **benign preneoplastic** lesions into **neoplastic cancer**. In this stage, due to increase in **DNA synthesis**, **cell proliferation** in the preneoplastic lesions, additional genotoxic events may occur resulting in additional DNA damage including **chromosomal aberrations and translocations**. These events result in the transfer from **preneoplastic**, clonally derived cell populations into **neoplastic cell** populations

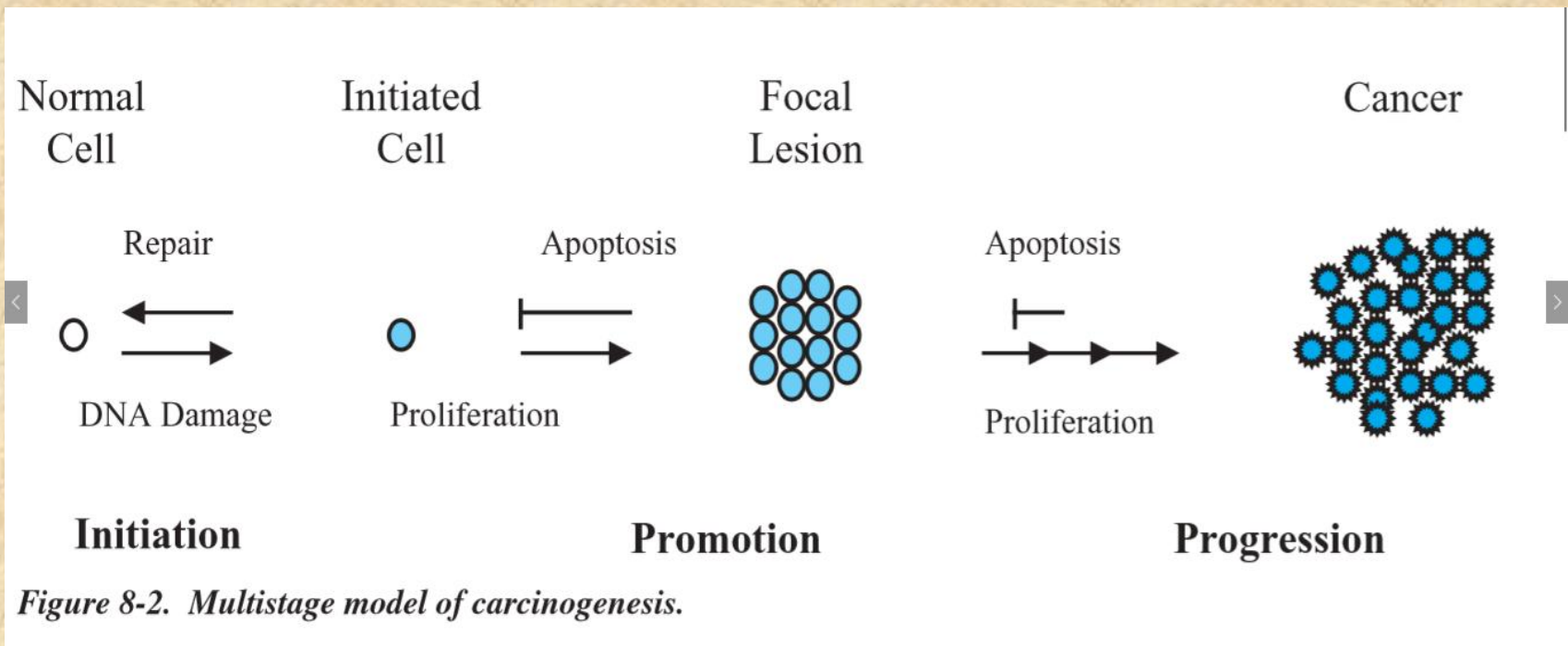


Figure 8-2. Multistage model of carcinogenesis.

MECHANISMS OF ACTION OF CHEMICAL CARCINOGENS

The development of neoplasia requires two major events: the formation of an **initiated, mutated cell** and the **selective proliferation** of the mutated cell to form aneoplasm

Genotoxic Carcinogens

Genotoxic carcinogens initiate tumors by producing **DNA damage**.

Examples of Genotoxic Carcinogens

Direct-acting carcinogens

- Nitrogen or sulfur mustards
- Propane sulfone
- Methyl methane sulfonate
- Ethyleneimine
- B-Propiolactone
- 1,2,3,4-Diepoxybutane
- Dimethyl sulfate
- Bis*-(Chloromethyl) ether
- Dimethylcarbonyl chloride

Chemicals requiring activation (indirect-acting carcinogens)

- Polycyclic aromatic hydrocarbons and heterocyclic aromatics
- Aromatic amines
- N*-Nitrosoamines
- Azo dyes
- Hydrazines
- Cycasin
- Safrole
- Chlorinated hydrocarbons
- Aflatoxin
- Mycotoxin
- Pyrrolizidine alkaloids
- Bracken fern
- Carbamates

Direct-Acting(Activation-Independent)Carcinogens

A variety of **carcinogens do not require metabolic activation** or chemical **modification to induce cancer**, and are termed direct acting or activation independent carcinogens. These chemicals are also defined as **ultimate carcinogens**. Examination of the chemical structure of these agents reveals that they are highly **reactive electrophilic** molecules that can interact with and bind to **nucleophiles**, such as cellular **macromolecules including DNA**.

Direct-acting carcinogens include epoxides, imines, alkyl and sulfate esters, and mustard gases

Direct-acting electrophilic carcinogenic chemicals typically **test positive** in the **Ames** test without additional bio activation with a liver metabolic fraction. The relative carcinogenic activity of direct-acting carcinogens is dependent upon such **competing reactions and also on detoxification reactions**.

Indirect-Acting Genotoxic Carcinogens

An important discovery in the understanding of chemical carcinogenesis came from **metabolic activation** to be carcinogenic. It has since been shown that the majority of DNA reactive carcinogens are found as **parent compounds**, or **procarcinogens**. **Procarcinogens** are stable chemicals that require subsequent metabolism to be carcinogenic.

The terms **procarcinogen**, **proximate carcinogen**, and **ultimate carcinogen** have been coined to define the **parent compound (procarcinogen)** and its metabolite form, either **intermediate (proximate carcinogen)** or **final (ultimate carcinogen)** that reacts with DNA. **The ultimate** form of the carcinogen is most likely the chemical species that results in mutation and neoplastic transformation.

Procarcinogen

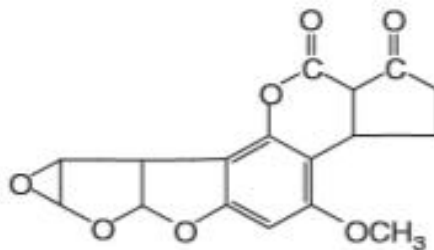
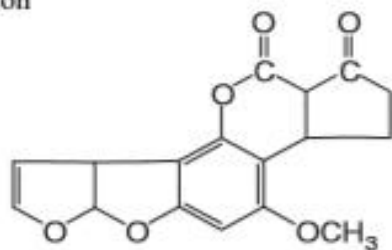


Proximate (Px) Carcinogen

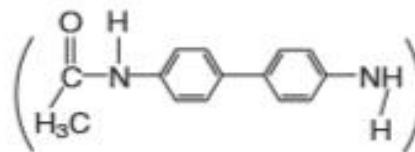
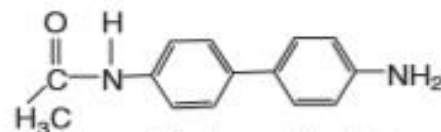
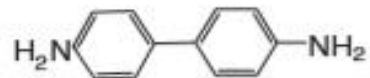


Ultimate (Ut) Carcinogen

Direct epoxidation



N-Hydroxylation

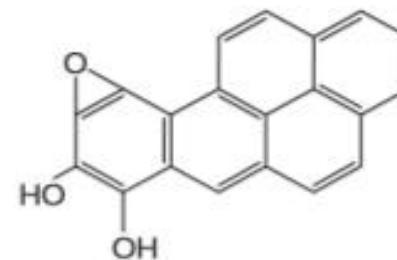
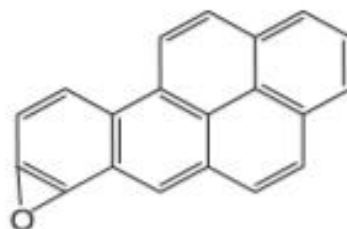


Benzidine (Pr)

N-hydroxy diacetyl Benzidine (Px)

N-Acetyl benzidine Nitrenium ion (Ut)

Two-step epoxidation



Benzo(a)pyrene (Pr)

Benzo(a)pyrene 7,8 epoxide (Px)

Benzo(a)pyrene 7,8 diol-9,10 epoxide (Ut)

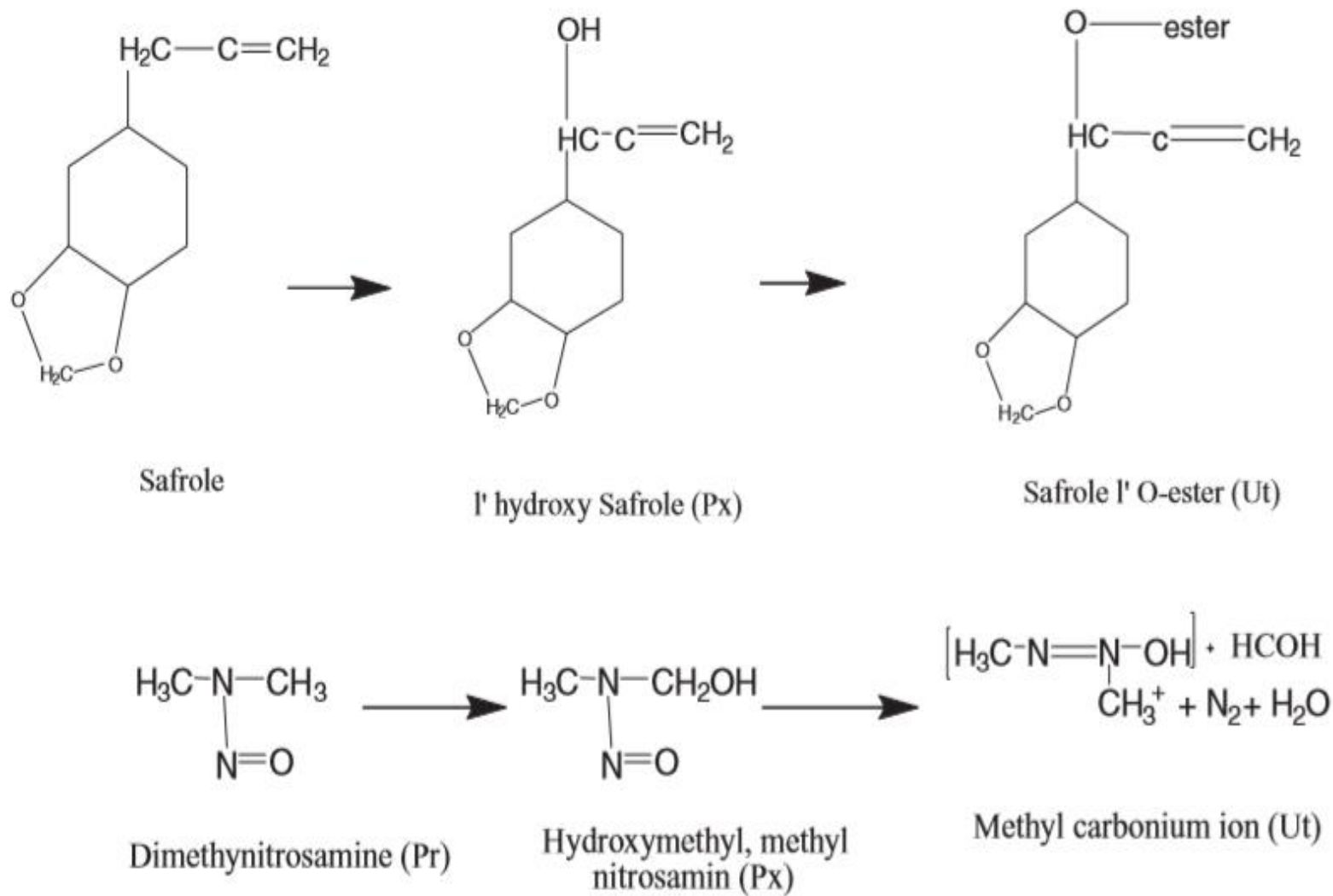


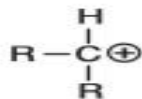
Figure 8-4. Structures of representative indirect-acting carcinogens and their metabolic derivatives.

The proximate (Px) and ultimate (Ut) carcinogenic forms result from the metabolism of the procarcinogenic form (Pr).

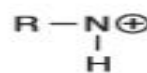
Damage by Alkylating Electrophiles

most chemical carcinogens require metabolic activation to exert a carcinogenic effect. The ultimate carcinogenic forms of these chemicals are frequently **strong electrophiles that can readily form covalent adducts with nucleophilic targets**. Because of their unpaired electrons, S:, O:, and N: atoms are nucleophilic targets of many electrophiles. The extent of adducts formed is limited by the structure of DNA, where **bulky electrophilic** chemicals can bind, and size of the ultimate carcinogenic form

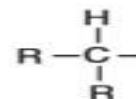
1. Carbonium ions



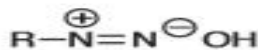
2. Nitrenium ions



3. Free radicals



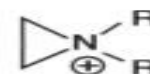
4. Diazonium ions



5. Epoxides



6. Aziridinium ions



7. Episulfonium ions



8. Strained Lactones



9. Sulfonates



10. Halo ethers



11. Enals

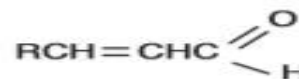
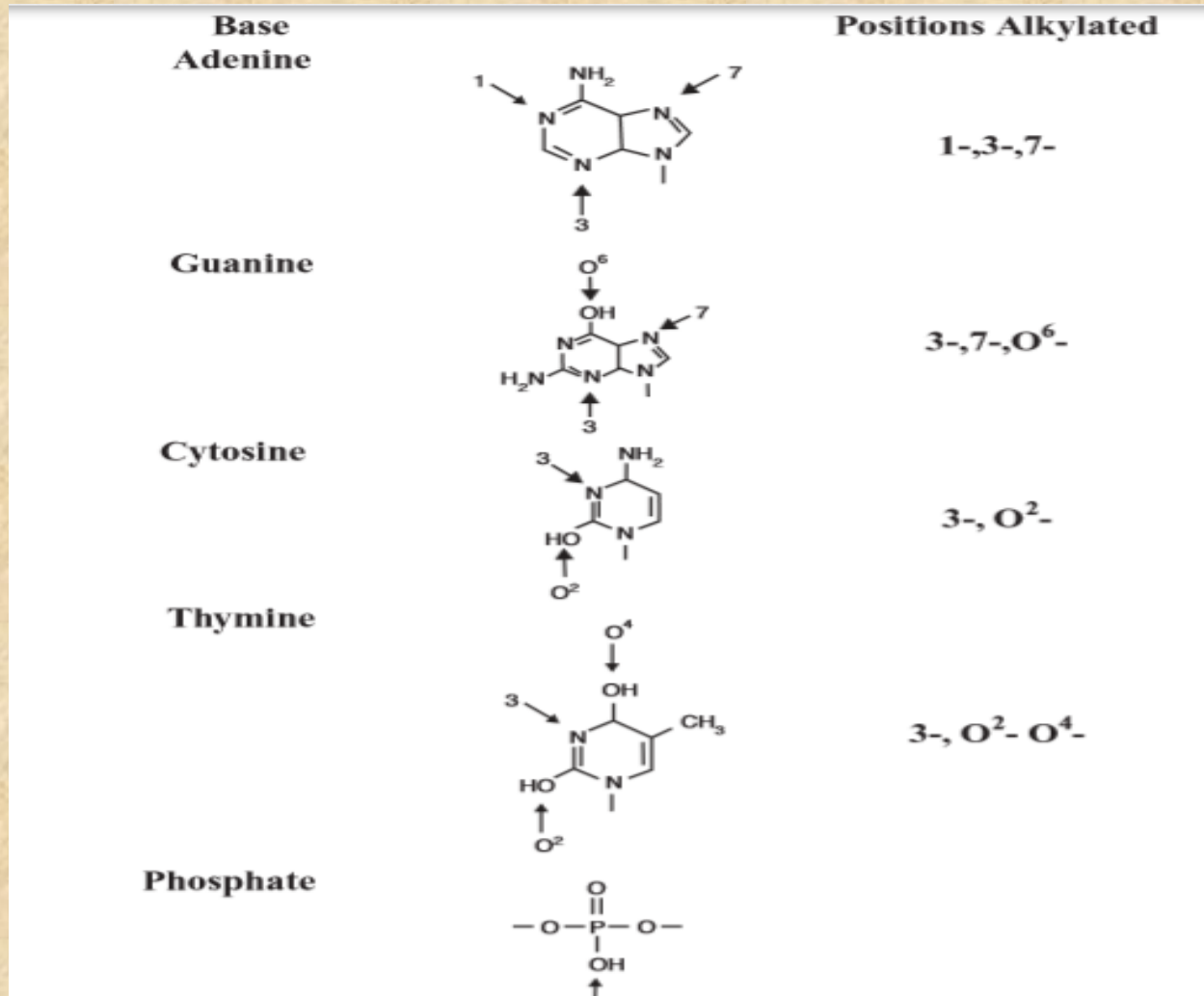
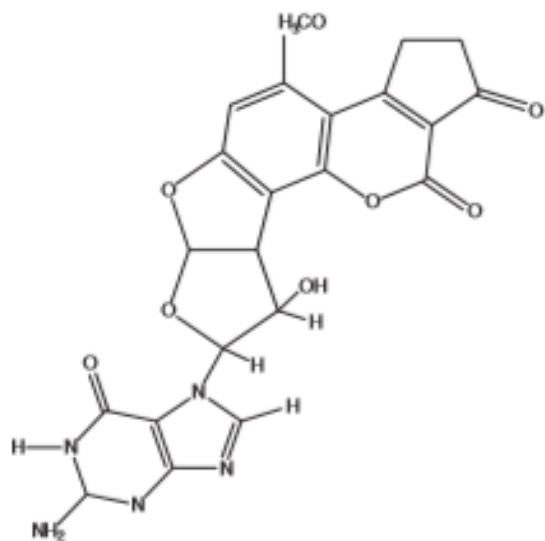


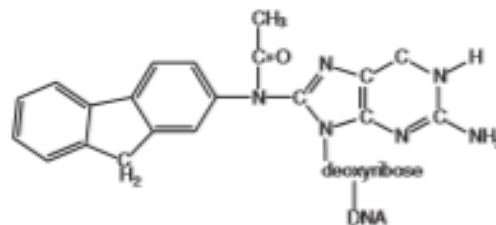
Figure 8-3. Structures of reactive carcinogenic electrophiles.

Where as carcinogen DNA adducts may be formed at all sites in DNA, the most common sites of alkylation include the **N7 of guanine, the N3 of adenine, the N1 of adenine, the N3 of guanine, and the O6 of guanine.** Alkylations of phosphate may also occur at a high frequency

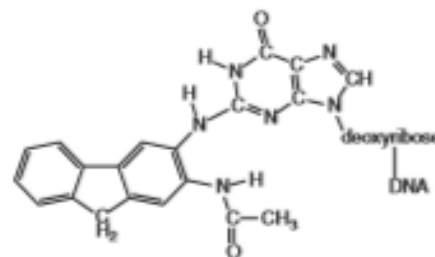




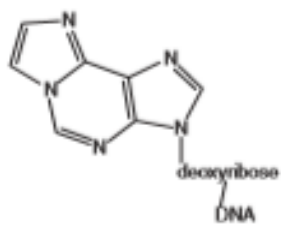
Aflatoxin B₁ N⁷-guanine-adduct



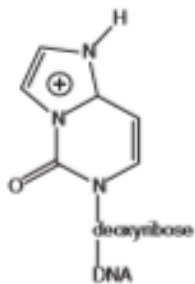
N-(deoxyguanosine-8-yl)-acetylaminofluorene in DNA



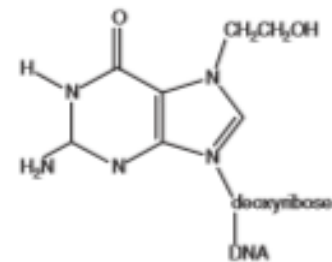
3-(deoxyguanosine N₂-yl)-acetylaminofluorene in DNA



1, N⁶-etheno-adenine in DNA



3, N⁴-etheno-cytosine in DNA



N⁷-(2-hydroxyethyl) guanosine in DNA

Figure 8-6. Select structures of protein and nucleic adducts of certain chemical carcinogens.

Different electrophilic carcinogens will often **display different preferences** for nucleophilic sites in DNA and different spectra of damage. **Dimethylnitrosamine and diethylnitrosamine**, for example, are metabolized by P450 oxidation to yield a **methyl carbonium ion (CH_3^+)** and an **ethyl carbonium ion (CH_3CH_2^+)**, respectively. Despite the structural similarities of the ultimate electrophiles.

The predominant **adduct** formed following exposure to methylating chemicals such as **methyl methane sulfonate is 7-methylguanine**.

Another common modification to DNA is the **hydroxylation of DNA bases**. **Oxidative DNA adducts** have been identified

The **source of oxidative DNA damage** is typically formed from free radical reactions that occur endogenously in the cell or from exogenous source.

Although a relatively large amount of **oxidative DNA adducts** have been proposed to be formed per day, **repair mechanisms** exist that maintain the cellular level at a low rate.

Methylation of deoxycytidine residues is a well-studied DNA adduct. This reaction occurs by the transfer of a **methyl** group from **S-adenosyl methionine** by **DNA methyl transferases**

Methylation of DNA results in heritable expression or repression of genes, with **hypomethylation** associated with active transcription of genes, while **hypermethylated** genes tend to be rarely transcribed.

Chemical carcinogens may **inhibit DNA methylation** by several mechanisms including forming **covalent adducts**, **single strand breaks** in the DNA, **alteration of methionine pools**, and **inactivation of the DNA methyltransferase** responsible for methylation

DNA Repair Mechanisms

Although cells possess mechanisms to repair many types of DNA damage, these are not always completely effective, and residual

DNA damage can lead to the **insertion of an incorrect base** during DNA replication, followed by **transcription and translation** of the mutated templates

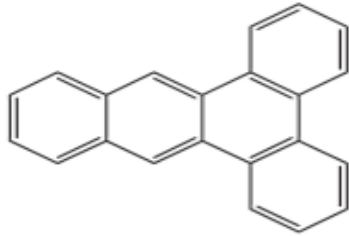
Table 8-8

DNA Repair Pathways

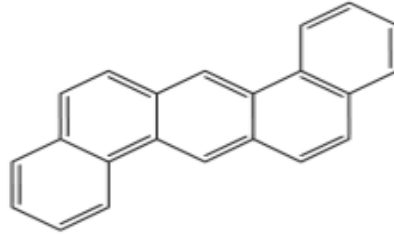
1. Direct reversal of DNA damage
2. Excision repair systems
 - Base excision repair
 - Nucleotide excision repair
 - Mismatch repair
3. Postreplicational repair (recombination repair)
4. Nonhomologous-end-joining (NHEJ): double-strand break repair

Classes of Genotoxic Carcinogens

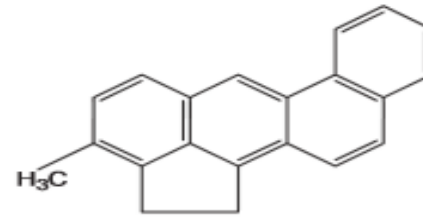
Polyaromatic Hydrocarbons are found at high levels in charcoal



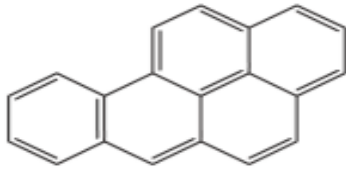
Dibenz(a,c)anthracene



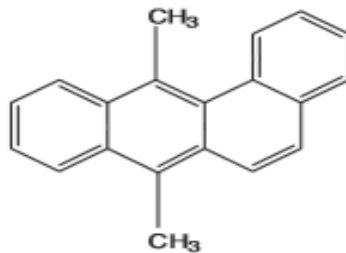
Dibenz(a,h)anthracene



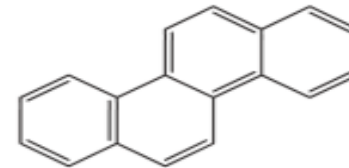
3-Methylcholanthrene



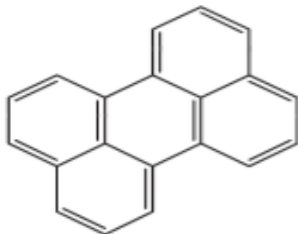
Benzo(a)pyrene



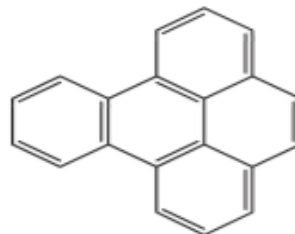
7,12-dimethylbenz(a)anthracene



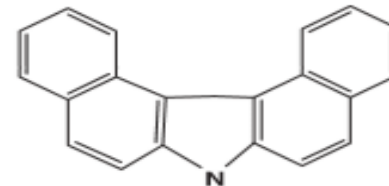
Chrysene



Perylene



Benzo(e)pyrene

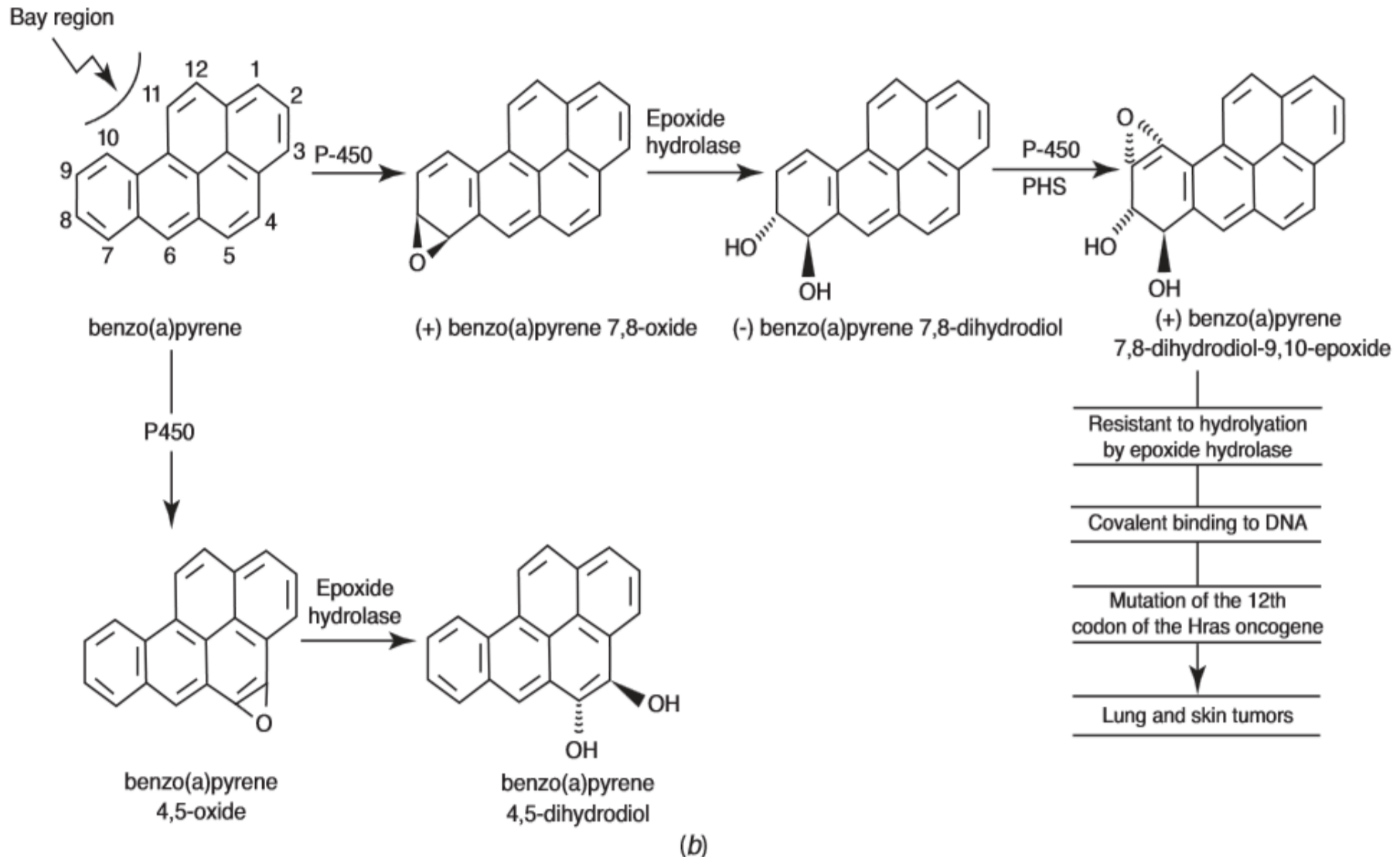


7H-Dibenzo(c,q)carbazole

(a)

(A) Chemical structures of selected carcinogenic polycyclic hydrocarbons

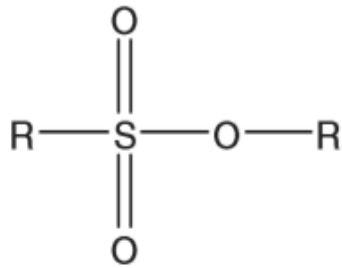
The metabolism and pathways that lead to tumor formation



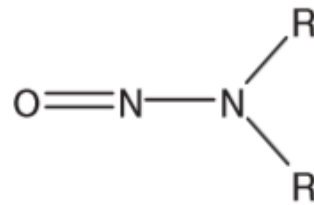
(B) Role of epoxide hydrolase in the activation of benzo[a]pyrene 4,5-oxide and in the conversion of benzo[a]pyrene to its tumorigenic bay-region diol epoxide.

Alkylating Agents Alkylating chemicals represent an important class of chemical carcinogens. Whereas some alkylating chemicals are direct-acting genotoxic agents, many **require metabolic activation to produce electrophilic metabolites** that can react with DNA. Alkylating agents can be classified into several groups including the **direct-acting alkylalkane sulfonate** the **indirect-acting nitrosamides**

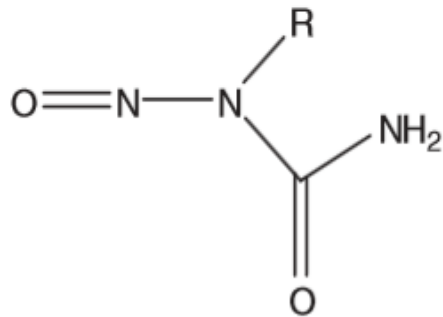
The alkylating compound diethyl nitrosamine and dimethyl nitrosamine readily react with DNA at more than 12 sites. **The N7 position of guanine and the N3 position of adenine are the most reactive sites in DNA for alkylating chemicals.** DNA methylation reactions occur more readily and thus exhibit >20 more adducts than with ethylation reactions. However, **ethylation** reactions have a greater affinity for oxygen centers, an event that appears to correlate with the mutagenicity and carcinogenicity of these compounds.



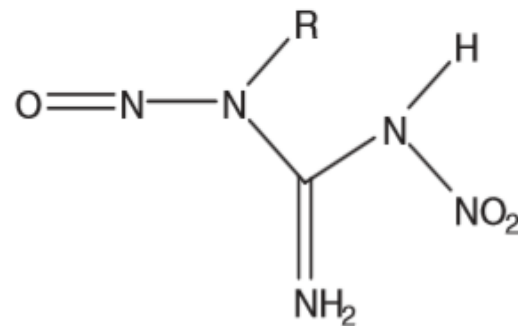
Alkyl alkanesulfonates



Diakyl nitrosamines



N-Nitrosoureas

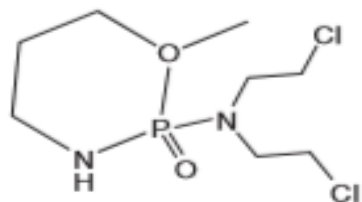


N-Alkyl-*N'*-nitro-*N*-nitrosoguanidine

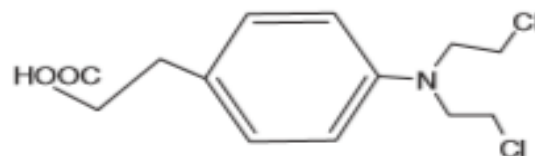
Figure 8-11. Structures of representative methylating and ethylating agents.

Alkalating chemicals including the nitrogen mustards (e.g., chlorambucil, cyclophosphamide) have been used in cancer chemotherapy. They produce DNA adducts as well as induce the formation of DNA strand breaks.

The alkylation of DNA by nitrogen mustards requires the formation of highly reactive N-alkyl aziridinium ions (Fig. below). Nitrogen mustards can produce a wide spectrum of mutations including base pair substitutions (AT and GC) and deletions. In addition, nitrogen mustards are causing chromosomal aberrations and sister chromatid exchanges



Cyclophosphamide



Chlorambucil

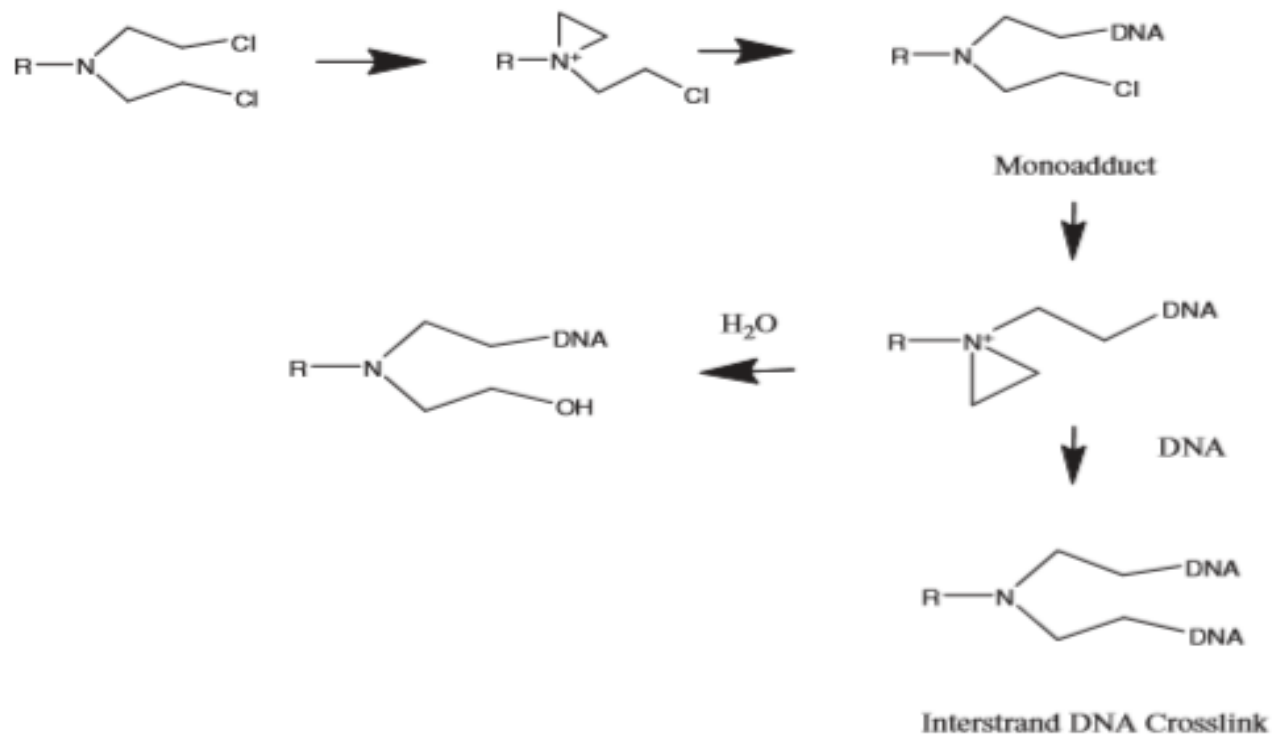


Figure 8-12. Nitrogen mustards and proposed mechanism for the reaction of nitrogen mustards with DNA.

Aromatic Amines and Amides

encompass a class of chemicals with varied structures (**aromatic amines, e.g. aniline dyes, 2-naphthylamine, benzidine, 2-acetylaminofluorene**) (Fig. below).

Because of their use in the dye industry and other industrial processes their carcinogen potential in humans was realized as early as the late 19th century. **While proper industrial hygiene processes have considerably reduced the human exposure to aromatic amines and amides in the workplace, exposure to these chemicals still occurs through cigarette smoke and environmental sources.** The aromatic amines undergo both **phase-I and phase-II** metabolism.

Phase-I reactions occur mainly by cytochrome P450-mediated reactions, yielding hydroxylated metabolites that are often associated with adduct formation in proteins and DNA, and produce **liver and bladder carcinogenicity**

For example, metabolism of **2-acetylaminofluorene** (AAF) results in the formation of N-hydroxy-AAF, which is a metabolite responsible for the liver tumorigenicity.

Similarly, **1-naphthylamine** exhibits carcinogenic activity only in test systems capable of producing the **N-hydroxy metabolite** of naphthylamine. **Aromatic amines are capable of forming adducts with several DNA bases.**

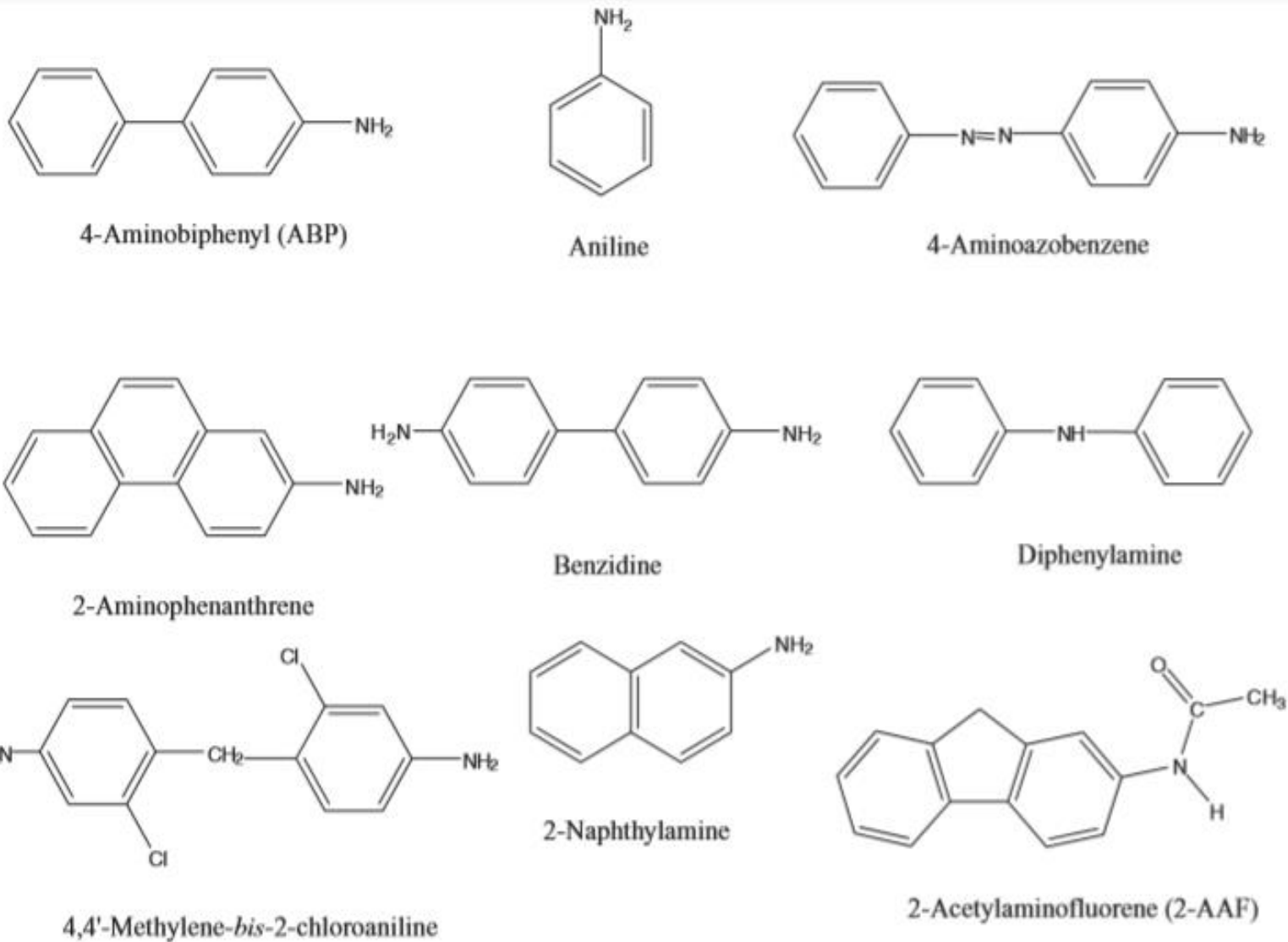


Figure 8-14. Chemical structures of selected carcinogenic aromatic amines.

Inorganic Carcinogens

Several metals exhibit carcinogenicity in experimental animals and/or exposed humans. Table a listing of some common metals and their corresponding carcinogenicity in animals and humans.

Carcinogenicity of Metals

METAL	ANIMAL			HUMAN	
	SPECIES	TUMOR SITE	TUMOR TYPE	EXPOSURE	TUMOR TYPE
Arsenic	Mice, dogs, rats	None observed	None observed	Cu refinery As pesticides Chemical plants Drinking water (oral)	Pulmonary carcinoma Lymphoma, leukemia Dermal carcinoma Hepatic angiosarcoma
Beryllium	Mice, rats, monkeys	Bone Lung	Osteosarcoma Carcinoma	None observed	None observed
Cadmium	Mice, rats, chickens	Injection site Testes	Sarcoma Teratoma	CD refinery	Pulmonary carcinoma
Chromium	Mice, rats, rabbits	Injection site Lung	Sarcoma Carcinoma	CR refinery Chrome plating Chromate pigments	Pulmonary carcinoma Gastrointestinal carcinoma
Cobalt	Rats, rabbits	Injection site	Sarcoma	None observed	None observed
Iron	Hamsters, mice, rats, rabbits	Injection site	Sarcoma	None observed	None observed
Lead	Mice, rats	Kidney	Carcinoma	None observed	None observed
Nickel	Mice, rats, cats, hamsters, rabbits Guinea pigs, rats	Injection site Lung Kidney	Carcinoma Carcinoma Carcinoma	Ni refinery	Pulmonary carcinoma Nasolaryngeal carcinoma Gastric and renal carcinoma Sarcoma (?)
Titanium	Rats	Injection site	Sarcoma	None observed	None observed
Zinc	Chickens, rats, hamsters	Testes Testes	Carcinoma Teratoma	None observed	None observed

Arsenic

Arsenic compounds are poorly mutagenic in both bacterial and mammalian cell assays .**Metallic arsenic, arsenic trioxide, sodium arsenite, sodium arsenate, potassium arsenite, lead arsenate, calcium arsenate, and pesticide mixtures** containing arsenic have been tested for carcinogenicity in experimental animals.

In contrast, inorganic arsenic compounds are known **human** carcinogens, based on sufficient evidence of carcinogenicity in humans. Inorganic arsenic compounds increases the risk of **cancer in the skin, lung, digestive tract, liver, bladder, kidney, and lymphatic and hematopoietic systems.**

The mechanisms for cancer formation are unclear but possibly involve the induction of **oxidative stress, altered cell signaling, modulation of apoptosis, and/or altered cell cycle.**

The latency period in humans of arsenic-related carcinogenesis is considered to be **30–50 years**. The first **signs of chronic** exposure, frequently seen in water supplies contaminated with arsenic, are skin pigmentation, depigmentation, hyperkeratosis of palms and soles, and skin lesions.

Cadmium and cadmium compounds have been classified as known human carcinogens based on evidence of carcinogenicity in humans, including epidemiological and mechanistic information that indicate a causal human cancer.

Many studies have shown that cadmium compounds **cause genetic damage, including gene mutations, DNA strand breaks, chromosomal damage, cell transformation, and disrupted DNA repair.**

cadmium exposure is associated with elevated **lung cancer** ,prostate cancer

Lead

Lead compounds also inhibits the activity of DNA and RNA polymerase in cell-free systems and in mammalian cell cultures.

soluble (lead acetate and lead subacetate) and **insoluble** (lead phosphate, lead chromate) **inorganic lead** compounds as well as for tetraethyl lead (an organic lead compound), following exposure via oral, injection, and in offspring exposed via the placenta or lactation.

Although **kidney tumors** (including adenomas, carcinomas, and adenocarcinomas) were most frequently associated with lead exposure, tumors of the **brain, hematopoietic system, and lung** were reported in some studies.