Health Effects of Air Pollution

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Air pollution has been distinguished on the basis of the chemical redox nature of its primary components.

was characterized by SO2 and smoke from incomplete combustion of coal accumulated during an inversion as a chilled,

Always had a characteristically "oxidant-type" pollution consisting of NOx and many secondary photochemical oxidants, such as O3, aldehydes, and hydrocarbon radicals.

Criteria Air Pollutants

- Gases: O_3 , CO, NO_x , So_x
- Particles: PM2.5, Pb

Exposure to Air Pollutants

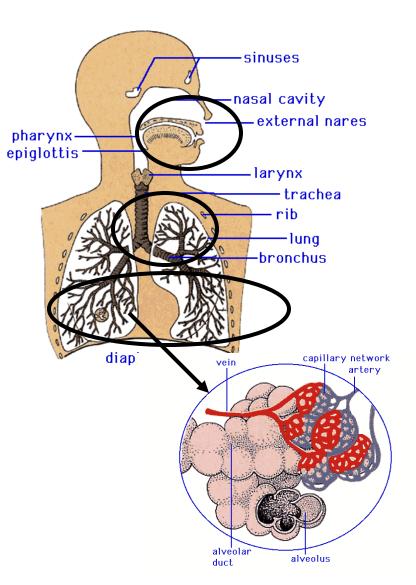
- Chronic
- Acute
 - Air Pollution Episode short term increase concentration
- Dependent on local conditions
- Epidemiological studies
 - Latency period
 - Lung cancer up to 30 years
- Toxicological studies
 - Determine effects of toxic substances
- Pollutant interactions



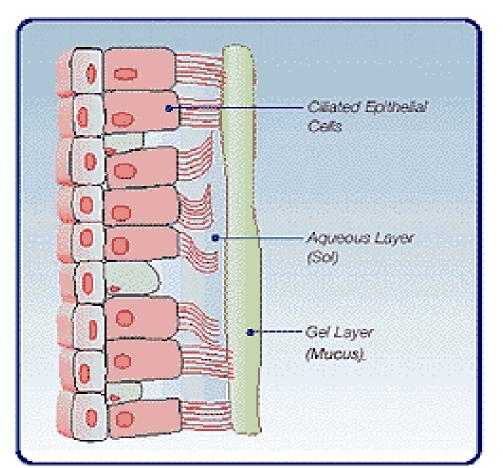
Smog Episode in New York City, 1963

Respiratory System

- Pollutants transported in via inhalation-respiratory tracts
- Person at rest breathes 12 to 15 times a minute (10 liters/min)
- 3 parts of respiratory system
 - Naso-pharyngeal (NAR)
 - Tracheo-bronchial (TBR)
 - Pulmonary-Alveolar Lungs serve as portal of entry
 - Highly permeable and lots of blood flow
 - Pulmonary-Alveolar Surface
 Area > 75 m²



Respiratory System



- Natural protection mechanisms (for particles)
- Naso-pharyngeal (HAR)
 - Nose hairs (filter particles)
 - Cough, Sneeze
 - Mouth breathing vs nasal breathing
- Tracheo-bronchial (TBR)
 - Mucociliary "escalator"
 - Bronchial constriction
- Pulmonary-Alveolar Macrophages (phagocytosis)
 - No cilia in Alveoli



Question: Do the natural protection mechanisms protect against toxic gases such as CO, O_3, SO_2 ?

Criteria Air Pollutants: Particulate Matter PM

- Small solid/liquid aerosol particles that remain suspended in air .
- Causes: materials handling, combustion processes, gas conversion reactions.
- Main sources: industrial processes, coal and oil burning, diesel motor vehicles

Criteria Air Pollutants: Particulate Matter

- Following inhalation: two possible fates
 - Deposition or Exhalation
- Particle Fate depends upon:
 - Aerodynamic & physiological behavior (human being)

Criteria Air Pollutants: Particulate Matter

- Wheezing & coughing to heart attacks and death
- TSP (Total Suspended Particles)
 - In presence of SO₂, direct correlation between TSP and hospital visits for bronchitis, asthma, emphysema, pneumonia, and cardiac disease
 - Studies suggest ~60,000 deaths from PM
 - 1% increase in mortality for 10 μ g/m³ increase in PM
 - Respiratory mortality up 3.4% for the same Cardiovascular mortality up 1.4% for the same

Criteria Air Pollutants: Carbon Monoxide CO

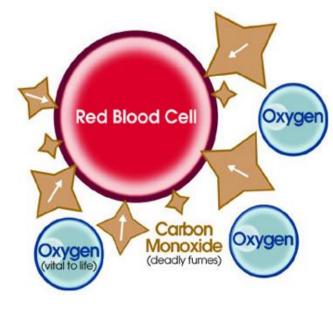
- Colorless, odorless, tasteless gas \rightarrow "Silent Killer"
 - Cause: incomplete combustion of carbon based fuels
 - Source: transportation sector, residential heating units
 - NAAQS regulates CO in outdoor air
- 50 ppm CO averaged over 8-hour period)

Criteria Air Pollutants: Carbon Monoxide CO

- Reacts with blood hemoglobin
 - Forms carboxyhemoglobin (HbCO) rather than oxyhemoglobin (HbO₂)
 - Prevents oxygen transfer

Toxic effects on humans

- Low-level: cardiovascular & neurobehavior
- Headaches/nausea/fatigue/ death
- Oxygen deficient to vulnerable people (anemia, chronic heart or lung disease, high altitude residents, smokers)
 - Cigarette smoke: 400-450 ppm; smoker's blood 5-10% HbCO vs 2% for non-smoker



Criteria Air Pollutants: Carbon Monoxide CO

- Concern in homes
- Install CO monitor
- No indoor CO regulations
 - >70 ppm \rightarrow flu-like symptoms (w/out fever)
 - 150-200 ppm → disorientation, drowsiness, vomiting
 - >300 ppm \rightarrow unconsciousness, brain damage, death
 - 500 Americans die/year from unintentional CO poisoning
- Treatment: fresh air, oxygen therapy, hyperbaric chamber



Criteria Air Pollutants: Ozone O₃

- Cause: atmospheric photochemical reaction
- Reactants: Hydrocarbons & Nitrogen Oxides
- NAAQS . 80 ppb 8 hr average
 - . 120 ppb 1 hr average
- Acute Health effects
 - Severe ear/nose/throat irritation
 - Eye irritation at 100 ppb ozone
 - Interferes with lung functions
 - Coughing at 2 ppm ozone
- Chronic Health Effects
 - Irreversible, accelerated lung damage
 - Why do we use ozone as disinfectant for water and wastewater treatment?



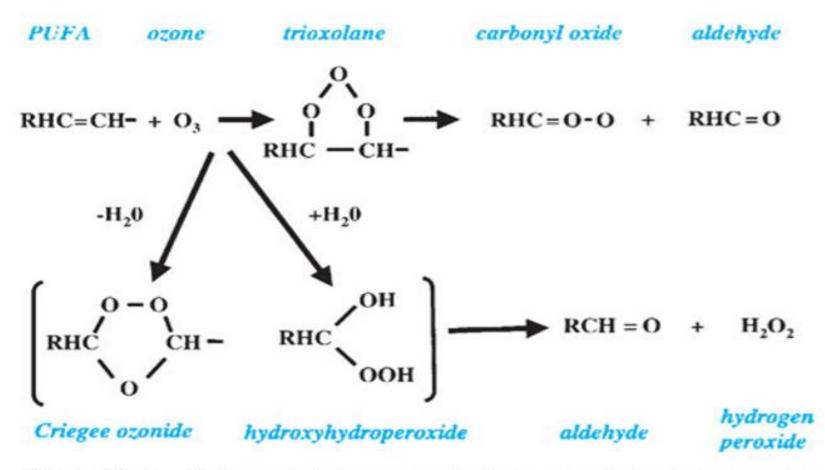


Figure 28-12. Major reactions pathways of O₃ with lipids in lung lining fluid and cell membranes. (Adapted with permission from the Air Quality Criteria Document for Ozone and Photochemical Oxidants. 600/P-93/004cF, NCEA Research Triangle Park, NC: U.S. EPA, 1996).

The mechanisms by which O3 causes injury have been studied using cellular as well as cell-free systems. As a powerful **oxidant,O3** seeks to extract electrons from other molecules.

The surface fluid lining the respiratory tract and cell membranes that underlie the lining fluid contain a significant quantity of polyunsaturated fatty acids (PUFA), either free or as part of the lipoprotein structures of the cell. The double bonds within these fatty acids have an, unpaired electron which is easily attacked by O3 to form ozonides that progress through a less stable zwitter ion or trioxolane(depending on the presence of water); these ultimately recombine or decompose to lipohydroperoxides, aldehydes, and hydrogen peroxide. These pathways are thought to initiate propagation of lipid radicals and autooxidation of cell membranes and macromolecule

Damage to the air-blood interface disrupts its barrier function and promotes inflammation. Inflammatory cytokines(e.g., interleukins 6, 8 and others, TNF, etc.) are released from epithelial cells and macrophages that mediate early responses and initiate repair. This inflammatory process is generally transient, but it may also **interact with neural irritant responses** to affect lung function acutely. This latter response may have implications for those with preexistent inflammation or disease

Criteria Air Pollutants: Nitrogen Oxides NO_x

- Cause: Fuel combustion at high temps
- Source: mobile & stationary combustion sources
- Prolonged exposure → pulmonary fibrosis, emphysema, and higher lower respiratory tract illness in children
- NAAQS = Annual Average 0.053 ppm as NO₂
- Toxic effects at 10-30 ppm NOx
 - Nose and eye irritation
 - Lung tissue damage
 - Pulmonary edema (swelling)
 - Bronchitis
 - Pneumonia
 - Aggravate existing heart disease

.Potential life-threatening exposure is are al-world problem for farmers, as near-lethal high levels of NO2 can be liberated from fermenting fresh silage.

Being heavier than air, the generated NO2 and CO2 displace air and oxygen and diffuse into closed spaces where workers can exposed to very high concentrations perhaps with depleted oxygen.

Typically, shortness of breath rapidly ensues with exposures nearing 75 to 100 ppm NO2, with delayed edema and symptoms of pulmonary damage.

Nitrogen dioxide is also an important indoor pollutant, especially in homes with unventilated gas stoves or kerosene heaters

Under such circumstances, very young children and their mothers who spend considerable time indoors may be especially at risk. studies indicate that NO2 is deposited along the length of the respiratory tree, with preferential deposition being in the distal airways.

the pattern of damage to the respiratory tract reflects this profile: damage is most apparent in the terminal bronchioles, more proximal in the airway than is seen with O3.

At high concentrations, the alveolar ducts and alveoli are also affected, showing their sensitivity to **oxidant** challenge. In the airways of these animals there is also damage to epithelial cells in the bronchioles, notably with loss of ciliated cells, as well as a loss of secretory granules in cells. The pattern of injury of NO2 is quite similar to that of O3, but its **potency** is about an order of magnitude lower

Criteria Air Pollutants: Sulfur Oxides SO_x

- Cause: Burning fuel that contains sulfur
- Source: Electric power generation, diesel trucks
- Soluble and absorbed by respiratory system
- NAAOS = 0.14 ppm 24 hr average
- Short-term intermittent exposures
 - Broncho-constriction (temporary breathing difficulty)
 - Ear/Nose/Throat irritation
 - Mucus secretion
- Long-term exposures
 - Respiratory illness
 - Aggravates existing heart disease

Criteria Air Pollutants: Lead (Pb)

- Source: burning fuels that contain lead (phased out), metal processing, waste incinerators, lead smelters, lead paint
- Absorbed into blood; similar to calcium
- NAAQS = $1.5 \ \mu g/m^3$ Pb Quarterly Average
- Accumulates in blood, bones, muscles, fat
 - Damages organs kidneys, liver, brain, reproductive system, bones (osteoporosis)
 - Brain and nervous system seizures, mental retardation, behavioral disorders, memory problems, mood changes,
 - Young children lower IQ, learning disabilities
 - Heart and blood high blood pressure and increased heart disease
 - Chronic poisoning possible



Criteria Air Pollutants: Air Quality Index (AQI)

- EPA (AQI) is for reporting daily air quality. The AQI focuses on short term health effects (1-48 hr after exposure). AQI is calculated from concentrations of SO₂, CO, O₃, and particles.
- AQI values in the 0-50 indicates Good air quality.
- AQI in the **51-100** range indicates Moderate air quality and exposures will cause short term health effects to some sensitive people (*and unhealthy effects for long-term exposure for most people*).
- Pilat opinion is that "Moderate" air quality is not very healthy.

Criteria Air Pollutants: Air Quality Index (AQI)

AQI is the highest magnitude of the PM,
 SO₂, CO, and O₃ individual Index values

AQI Value	Air Quality	24 hr PM2.5	24 hr SO ₂	8 hr CO	8 hr O ₃
		(µ g/m³)	(ppm)	(ppm)	(ppm)
0-50	Good	0-15.4	0.0034	0.0-4.4	.000064
51-100	Moderate	15.5-40.4	.035144	4.5-9.4	.065084
101-150	Unhealthy to Sensitive	40.5-65.4	.145224	9.5-12.4	.085104
151-200	Unhealthy	65.5-150.4	.225304	12.5-15.4	.105124
201-300	Very Unhealthy	150.5-250.4	.305604	15.5-30.4	.125374
NAAQS		35 µg/m ³	0.14 ppm	9 ppm	.08 ppm

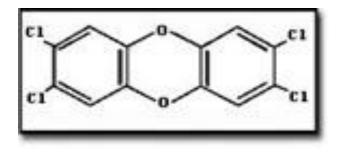
Hazardous Air Pollutant HAP: Mercury Hg

- Elemental Hg inhaled as a vapor, absorbed by lungs
- Cause: vaporized mercury
- Sources: coal combustion, accidental spill, mining (*teeth silver fillings*)
- Effects: Nervous system (acute, high), respiratory system (chronic, low), kidneys, skin, eyes, immune system; Mutagenic properties

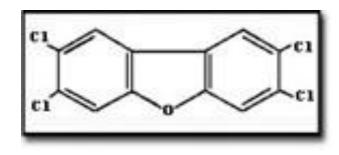
- Symptoms
 - Acute: chills, nausea, chest pains/tightness, cough, gingivitis, general malaise
 - Chronic: weakness, fatigue, weight loss, tremor, behavioral changes

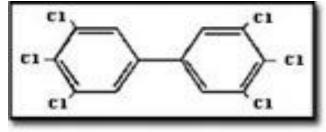
Hazardous Air Pollutant Dioxins

- Generic term for several chemicals that are highly persistent in the environment
 - chlorinated dibenzo-p-dioxins (CDDs)
 - chlorinated dibenzofurans (CDFs)
 - certain polychlorinated biphenyls (PCBs)
- Cause: burning hydrocarbons in presence of chlorine or chlorides
- Sources: waste incinerators
- Notice the Chlorine atoms on the benzene ring type molecules (probably all these type^{2,3,7,8}-Tetrachlorodibenzo<u>furan</u> compounds are carcinogenic)
- Does using Chlorine to treat drinking water result in the formation of carcinogenic compounds?



2,3,7,8-Tetrachlorodibenzo-p-dioxin





3,3',4,4',5,5'-Hexachlorobiphenyl

Hazardous Air Pollutant: Dioxins



Comparative Photos Showing Immediately Prior To And Immediately Following Dioxin Poisoning

(Note: this is an extreme case of dioxin poisoning)

- Varying toxicity
 - Problems with high exposures
 - Exact effects of low exposures not known
- Health Effects
 - Carcinogenic
 - Some are "known human carcinogen" (2,3,7,8 tetrachlordibenzo-p-dioxin, TCDD)
 - Other dioxins are "reasonably anticipated to be a Human Carcinogen"
 - Reproductive and developmental effects
 - Chloracne

Aldehydes Carbonyl compounds, notably short-chained (2–4 C) aldehydes, are common photo-oxidation products of unsaturated hydrocarbons.

Two aldehydes are of major interest by virtue of their concentrations and irritancy:

formaldehyde (HCHO) and a crolein (H2C CHCHO). They contribute to the odor as well as eye and sensory effects of smog.

Formaldehyde accounts for about 50% of the estimated total aldehydes in polluted air, while acrolein, accounts for about 5% of the total

Formaldehyde and particularly a crolein are also found in main stream tobacco smoke .Formaldehyde is also an important indoor air pollutant and can often achieve higher concentrations indoors than outdoors due to out-gassing by new upholstery or other furnishing

Formaldehyde is a primary sensory irritant. Because it is very soluble in water, it is absorbed in mucous membranes in the nose, upper respiratory tract, and eyes Along time concern regarding formaldehyde has been its potential carcinogenicity by the detection of DNA adducts

Nasal cancer had been induced empirically with formaldehyde vapor in a 2-year study where rats were exposed to 2, 6, or 14 ppm 6 hours per day, 5 days per week.

Other Aerosols: Bioaerosols

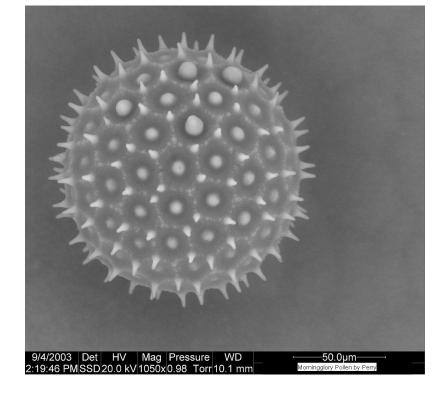


Mechanical aeration at Waste water treatment plant

- Aerosols with organic origin
 - Non-viable: pollen, dander, insect excreta, sea salt
 - Viable: microorganisms
- Cause: aerosolization of organic materials
- Sources:
 - Human: sneezing, coughing
 - wind, waves, Waste water treatment plants, cooling towers
 - Health Effects: allergies (pollen) to death (pathogenic organisms)
 - Pathogenic Minimum Infectious Dose

Other Aerosols: Bioaerosols

- Allergies
 - Pollen, dander, fungi (spores)
- Airborne transmission of disease
 - Bird flu,, Legionnella (pneumonia)
 - Indoor Air Quality
 - Ventilation Systems moist ductwork, protection, recycled air
 - Office Buildings Sick Building Syndrome
 - Hospital (nosocomial)
 - Biological Warfare
 - Anthrax, Ebola virus



Morning Glory Pollen SEM