



# Purification and characterization of tannase from the local isolate of *Aspergillus niger*

Saja Taha Yaseen Al-Mraai, Dhia Falih Al-Fekaiki, Alaa Jabbar Abd Al-Manhel\*

Department of Food Sciences, College of Agriculture, University of Basrah, Basrah City, Iraq.

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## ABSTRACT

Fungi of genus *Aspergillus* are active producers of extracellular tannase, and characteristics of enzyme production by local *Aspergillus* species have been well studied. The aim of this experimental study is isolation and partial purification of tannase enzyme from the *Aspergillus niger* fungus species. The purification was achieved by applying respectively ammonium sulfate (50-70%), dialysis with a specific activity 1227.08 (unit/mg protein), then passed the enzymatic extract through superdex 200 column in the gel filtration chromatography by using ÄKTA Pure 25 apparatus, the activity and the specific activity reached to 525.12 (unit/ml) and 2917.33 (unit/mg protein) respectively, with purification fold of 11.629 and yield 18.40%. One peak was obtained from an enzyme purity, which diagnosed by the absence of denaturation substances for protein SDS. Characteristics of pure enzyme showed that the molecular weight of the pure enzyme was determined, and it was found that the weight of enzyme was 95.49 KDa by using sodium dodecyl sulfate (SDS) polyacrylamide gel electrophoresis, the enzyme was active in pH range 3 to 6 and temperature of 20 to 50°C. The optimum pH and the temperature for tannase activity were recorded to be pH 5 and 35°C, respectively. Moreover, the tannase activity was inhibited by Cu, Fe and Hg ions while increased by Mg and K ions when added at 1 and 5 mM.

## 1. INTRODUCTION

Tannase or tannin acyl hydrolase (E.C: 3.1.1.20) is an enzyme works on the hydrolysis of tannic acid through breaking the ester and side bonds to its primary units of glucose and gallic acid. The enzyme is important industrially and has several applications in various industries such as food, animal feed, cosmetics, pharmaceutical, chemical, and leather industries [1]. Many foods contain a high level of tannin including tea, coffee, plenty of spices and some types of fruit and vegetables like grapes, apricots, dates, pomegranates strawberries, eggplant as well as nuts, such as peanuts and nuts [2]. Studies have shown that the tannin is responsible for reducing the amount of intaking food, in other mean an anti-nutrition factors, as a result of lowering the food efficiency and the net energy of metabolism and reduce the digestive protein because of its effect on the digestive enzyme, then lowering digestion of an amino acids as well as preventing degradation of the starch inside alimentary canal and cause damage to the cells lining the stomach and thus influence in the degree of absorption of the rest food elements and its ability on the linkage with the protein to form complex protein-

tannin or precipitate in the form of a gel in water solutions, so the foods which have a high levels of tannin regard a low nutritional value food, as well as tannin, has a negative impact on the iron absorption being associated with iron in the digestive system thereby reducing of its vital provided and absorption [3,4,5]. The current study aims to purification and characterization of tannase produced from a local fungal isolate for the purpose of using its in clarification of juices and reduce the content of tannin then removing the bitterness in beverages, juices, and tea, also the gallic acid producing from degradation by added tannase has an importance like antioxidant and antimicrobial activity which is more positive on the quality of the juices.

## 2. MATERIALS AND METHODS

### 2.1. Isolation, Production, and Extraction

Forty fungal isolates of molds were locally isolated from soil and tea waste then isolates molds were diagnosed at Biotechnology lab. Of Agriculture College-University of Basrah-Iraq, *Aspergillus niger* was selected as potential extracellular tannase depending on the phenotypic and microscopic characteristics of colonies accordance with the taxonomy keys that mentioned in [6,7]. Wheat Bran moistens with the pomegranate peel extract (1:2) was used for tannase production and enzyme extracted with acetate buffer (50 ml/10 g, pH 5, 0.2 M) at 200 rpm for 1 hr [8].

\*Corresponding Author

Alaa Jabbar Abd Al-Manhel, Department of Food Sciences, College of Agriculture, University of Basrah, Basrah City, Iraq.

E-mail: [alaafod\\_13@yahoo.com](mailto:alaafod_13@yahoo.com)

## 2.2. Purification of Enzyme

### 2.2.1. Concentration by Ammonium Sulfate

Crystals of ammonium sulfate were added to the crude extracted gradually for the concentration of the enzyme with stirring continuously for 4 hours until solubilization with saturation ratio between 20-80%. Centrifugation at 10000 rpm for 15 min. at 4°C. In each step estimated the enzymatic activity and concentration of protein in the filtrate and precipitate [9].

### 2.2.2. Enzyme Assay

Enzymatic activity was estimated according to [10]. One unit of tannase activity was defined as the amount of enzyme that hydrolyzed 1 Microgram of tannic acid in 1 minute under the assay condition. The specific activity of an enzyme is calculated by dividing the enzyme activity on protein concentration.

### 2.2.3. Protein Assay

A concentration of protein was measured according to [11].

### 2.2.4. Gel Filtration

The process of gel filtration was done by using ÄKTA Pure 25 apparatus with Superdex 200 10/300 GL column (10 mm diameter, column size 23.562 ml, pressure 1.5 mpa, dimensions 10 × 300 mm, flow rate of 0.5 ml/min) filled packaged with agarose and dextran, the column was calibrated with acetate buffer pH 5 (4 liters), then injecting the concentrated sample (0.5 ml) which was obtained from the precipitation of ammonium sulphate in the column gradually after filtered by millipore filter 0.22 µm and following the separated peaks on the wavelength of 280 nm, then measured the enzymatic activity (unit/ml).

### 2.2.5. Electrophoresis

The purification of an enzyme was estimated by using a polyacrylamide gel electrophoresis with an absence of the denaturation substances sodium dodecyl sulfate (SDS).

## 2.3. Characterization of the Purified Enzyme

### 2.3.1. Determination of molecular weight of enzyme

Polyacrylamide gel electrophoresis with a presence of SDS was performed according to the method of [12] modified by [13] using the slab electrophoresis apparatus. The molecular weight was determined from the relative mobility (calculated by measuring the amount of movement of the protein and dividing that by the distance of the movement dye (bromophenol blue) from the origin) of the standard protein such as glucose oxidase (160 KDa), bovine serum

albumin (67 KDa), papain (45 KDa), pepsin (34 KDa) and lysozyme (14.4) KDa.

### 2.3.2. Optimum pH for Enzyme Activity and Stability

Sodium acetate buffer of pH ranges from (3-5.5) and phosphate buffer of pH ranges from (6-8) with a concentration of 0.2 M were used to calculate the optimum pH for enzymatic activity (unit/ml), while the optimum pH for stability was examined as remaining activity (%) using different buffer solutions of pH ranges between (3-9) at 30°C for 2 h using the tannin as a substrate.

### 2.3.3. Optimum Temperature for Enzyme Activity and Stability

The enzyme was incubated in a water bath to estimation the optimum temperatures with a range between (20-80)°C for 30 min., where the optimum temperature for stability was estimated after incubation tannase in a water bath at temperatures ranged from (20-80)°C for 1 h, then cooling directly in an ice bath.

### 2.3.4. Effect of metal ions on enzyme activity

Equal volumes of purified enzyme were mixed with 1 and 5 mM of metal salt (MgCl<sub>2</sub>, HgCl<sub>2</sub>, FeSO<sub>4</sub>, ZnSO<sub>4</sub>·7H<sub>2</sub>O, KCl, CuCl<sub>2</sub>·2H<sub>2</sub>O, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, CaCl<sub>2</sub>) then incubated at 30°C for 60 min after that enzyme activity was assayed and remaining activity was calculated.

## 3. RESULTS AND DISCUSSION

### 3.1. Isolation

40 Pure isolation isolated from different sources as soil belongs to 7 genes, it was observed the superiority of *Aspergillus niger* in the production of tannase. Noticing [14] when isolating 35 molds isolation near from desert plants that 64.57% isolates were produced tannase, while [15] was able to isolate 16 molds isolation producing for tannase from different parts of plants.

### 3.2. Purification

The specific activity was increasing gradually until reaching to 1227.08 (unit/mg) at saturation of 50-70% with purification fold of 4.891 and yield 50.821% as shown in Table 1. The obtained results agreed with many of the studies that pointed to the importance of the use of ammonium sulfate as the first step of tannase purification [16] was used of ammonium sulfate at saturation rate 70% to concentration of the enzyme with a specific activity 4.58 (unit/mg) with purification fold of 1.91 and yield 68.75%, while [17] were showed the possibility of enzyme precipitation by ammonium sulfate with saturation 75% from *Aspergillus niger* with enzymatic yield 72.5% and purification fold of 1.4.

Table 1: Purification of tannase from *Aspergillus niger*.

Steps of purification	Volume (ml)	Activity (unit/ml)	Protein (mg/ml)	Specific activity (unit/mg)	Total activity (unit)	Purification	Recovery %
Crude extract	200	228.29	0.91	250.86	45658	1	100
Precipitation of (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> (50-70)%	31	748.52	0.61	1227.08	23204.12	4.891	50.821
Gel filtration Superdex 200	16	525.12	0.18	2917.33	8401.92	11.629	18.40

### 3.3. Gel Filtration

The results shown in Figure 1 an existence of a two peaks of the enzymatic extract after precipitation by ammonium sulphate and measured the activity of protein on the wavelength of 280 nm, where appeared a high enzymatic activity in the first peak 525.12 (unit/ml), which was chosen to complete the study with enzymatic yield 18.40% and purification fold of 11.629. The obtained results were similar to

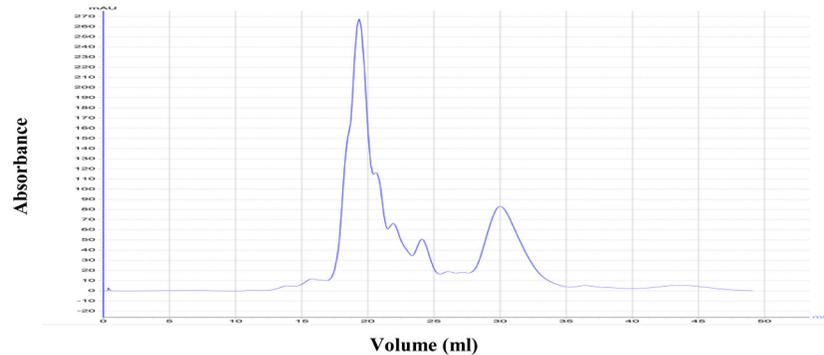


Figure 1: Chromatogram of gel filtration for tannase enzyme by ÄKTA Pure 25 using Superdex 200 10/300 GL.

### 3.4. Electrophoresis

It was noted that there were 6 protein bands (track A) for the crude extracted in polyacrylamide gel with an absence of denaturation materials of protein SDS, while appeared one protein band (track B) for enzyme after gel filtration, which means a high purity of enzyme as well as enhancing the results of gel filtration step as shown in Figure 2, seeing that [19] had gotten a one protein band represented the pure enzyme from *Paecilomyces variotii*.

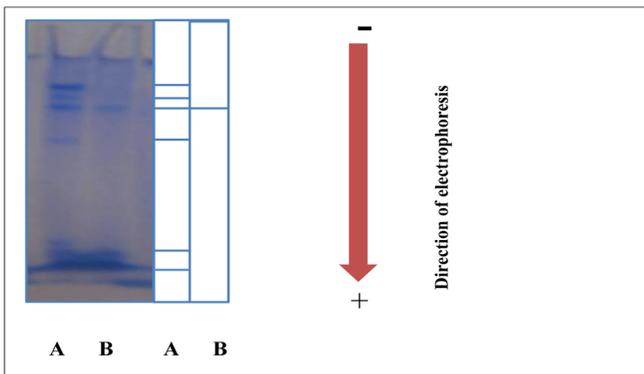


Figure 2: Determination of tannase purity by electrophoresis in polyacrylamide gel without SDS, A: crude extracted; B: pure enzyme after gel filtration.

### 3.5. Characterization of Enzyme

#### 3.5.1. Molecular Weight

Estimate the molecular weight for tannase by SDS-PAGE in the gel of a polyacrylamide using standard proteins as shown in Figure 3, that explains the relationship between logarithm the molecular weight of standard protein and relative movement of the protein itself, which shows that the molecular weight of tannase reached to 95.49 KDa.

Figure 4 explains the process of electrophoresis in polyacrylamide gel with SDS for standard proteins and pure enzyme, the results were

what many researchers have found where in made [18] a purification of the enzyme with a gel filtration technique using sephadex G-200 with a number of purification fold of 12.25 which produced from a fungus *Mucor* spp. also, the results agreed with [16] when studying enzyme purification of *Aspergillus niger* which purified by gel filtration technique using sephadex G-75 with an enzymatic yield 24% and purification fold of 2.5.

different between researchers, it agreed with the findings of [20] that the molecular weight for tannase product from *Aspergillus niger* reached to 89.9 KDa estimated in an SDS-PAGE, while molecular weight was 80 KDa for tannase product from *Aspergillus niger* in SDS-PAGE [16].

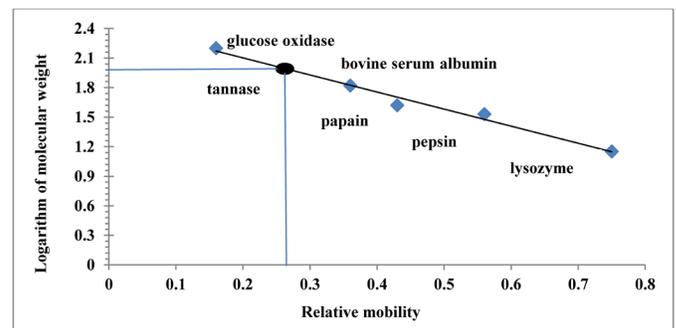


Figure 3: Standard curve to determine the molecular weight of tannase produce from *Aspergillus niger* ( $A_{13}$ ) by SDS-PAGE.

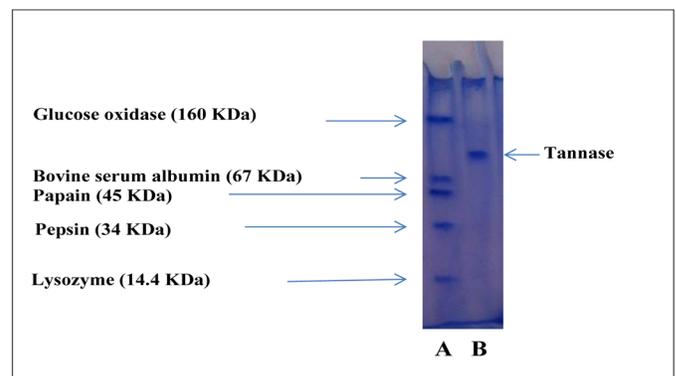


Figure 4: Electrophoresis with SDS in polyacrylamide gel, A: standard proteins; B: pure tannase.

### 3.5.2. Optimum pH for Activity and Stability

Appeared from the results in Figure 5 that the optimum pH for the enzyme activity is 5 with enzymatic activity reached 525.2 (unit/ml), the results of the current study be in conformity with [21] that the optimum pH for tannase activity which produced from *Aspergillus tamaraii* is 5, also [22] noted that the optimum pH of the tannase activity was 5, as well as [23] clarify that the optimum pH of the tannase activity for free and restricted enzyme were 5 and 5.5 respectively, as referred [24] that the optimum pH for tannase activity was 5.5, while the results of the optimum pH for tannase stability produced from the local isolation *Aspergillus niger* ( $A_{13}$ ) ranged between (3-6) as shown in Figure 6 where the enzyme had kept most of its activity for an 2 hours of incubation, while observed decreasing in pH at the values of 8, 8.5 and 9 where the remaining activity was 15, 20 and 25% respectively, this decline is attributed to the influence of the pH in change the secondary and tertiary structure of the enzyme molecule and change the form of active site [25]. The results were similar to the findings of many researchers, so [22] pointed out that the optimum pH for tannase produced of fungi ranged between (4.5-6) also clarify [24] that the optimum pH for tannase stability ranged between (3-6), either [26], explained that the optimum pH is 7.4 for tannase stability.

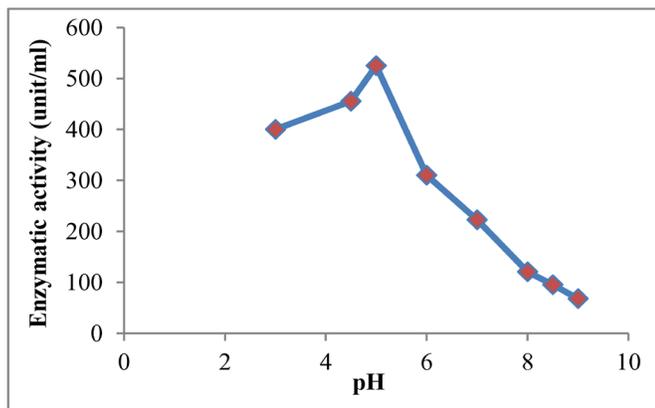


Figure 5: Optimum pH for tannase activity.

(unit/ml) at a temperature of 50°C and continued to decline until reaching to 25.7 (unit/ml) at a temperature of 80°C. The result has agreed with the result of [22] in determining the optimum temperature of tannase activity which was 35°C, while noted [24] through the study of the optimum temperature of tannase activity was 60°C, as well as determined [27] that optimum temperature of pure tannase from *Aspergillus tamaraii* was 30°C for the free enzyme and 45°C for the restricted enzyme. Tannase was founded thermally stable in a range between (30-50)°C as kept of its fully activity when incubation at these grades of temperature for 60 minutes, then lowered the activity rapidly with rising temperature even lost its full activity at 90°C, while an enzyme kept about 50% and 38.4% at a temperature of 60°C and 70°C respectively as clarified in Figure 8. The results were similar to findings of other researchers where reach [22] to determine the optimum temperature for enzyme stability, which ranged between (25-45)°C, as well as [24] found that tannase was thermally stable at temperatures of (40-60)°C, as mentioned [27] that pure tannase from *Aspergillus tamaraii* with its forms free and restricted enzyme was kept of its activity at temperatures ranging from (30-45)°C.

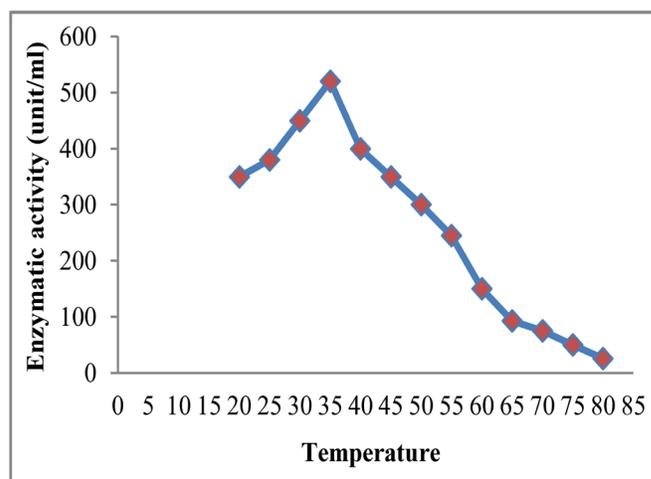


Figure 7: The optimum temperature for tannase activity.

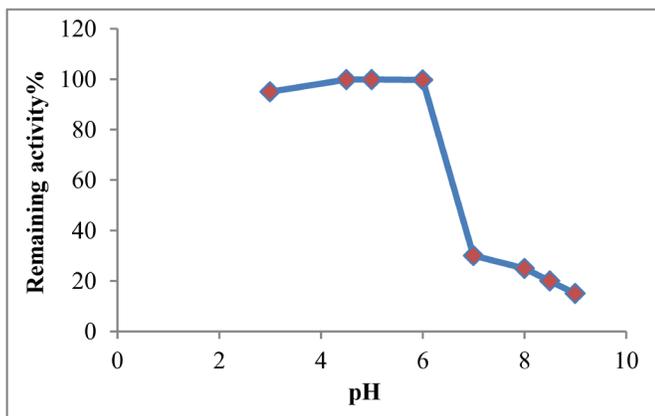


Figure 6: Optimum pH for tannase stability.

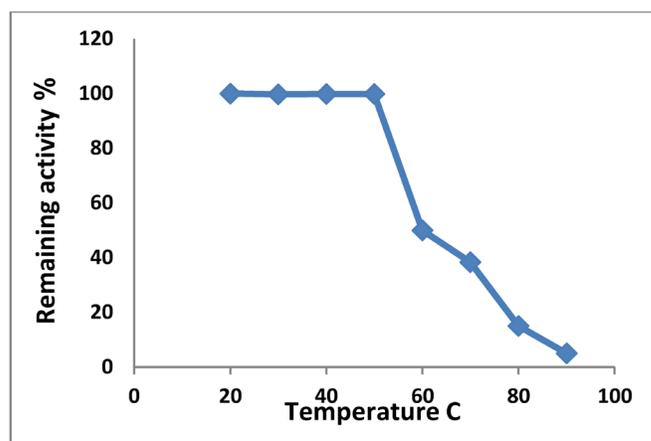


Figure 8: The optimum temperature for tannase stability.

### 3.5.3. The Optimum Temperature for Activity and Stability

Observe from Figure 7 increasing the enzymatic activity with increasing the temperature until reached to its maximum of 520.5 (unit/ml) at a temperature of 35°C, then decrease gradually to 250

### 3.5.4. Effect of Metal Ions on Enzyme Activity

The results of tannase treated with metal ions in Table 2 showed that the tannase was increased when treated with 1 mM of  $MgCl_2$  and KCl

while Zn and Ca slightly decreased of tannase activity at the same concentration. Some metal ions bind to the tannase and alter the enzymatic activity by stabilization or destabilization of the enzyme's conformation [28]. Furthermore, Cu, Fe, and Hg decreased in tannase activity occurred after incubation with 1 and 5 mM. The inhibition of tannase activity by these ions may indicate the presence of -SH group in the active site of the enzyme leading to oxidize theme moreover it may form a complex with the enzyme which prevents it from binding to the substrate and form the product. These results agree with [27] who showed the Mg ion was considered as activated for the enzyme, but Hg, Ca, Fe, Zn and Cu were an inhibitor of tannase activity.

**Table 2:** Effect of metal ions on tannase activity produces from *Aspergillus niger* (A<sub>13</sub>).

Metal Salts	Concentration (Mm)	Remaining activity (%)
Control	0	100
MgCl <sub>2</sub>	1	106
	5	100.51
HgCl <sub>2</sub>	1	25.12
	5	22.32
FeSO <sub>4</sub>	1	30.02
	5	28.04
ZnSO <sub>4</sub> ·7H <sub>2</sub> O	1	92.82
	5	87.3
KCl	1	100.45
	5	99.8
CuCl <sub>2</sub> ·2H <sub>2</sub> O	1	32.16
	5	30.71
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	1	73.24
	5	71.30
CaCl <sub>2</sub>	1	85.36
	5	70.1

#### 4. CONCLUSION

In conclusion of the current study, *Aspergillus niger* a local isolate was found to be able to secrete tannase by SSF using wheat bran agroindustrial waste as a substrate for the production of this enzyme high industrial relevance, especially for the food industry. Tannase was purified from culture supernatant by ammonium sulfate, dialysis and gel filtration with AKTA pure a new technique, However, both the purification and the characterization of the enzyme are necessary to determine the commercial viability of the enzyme and possible application for industrial purposes, such as juice, tea, cosmetic, etc.

#### 5. ACKNOWLEDGMENT

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