

Gas exchange

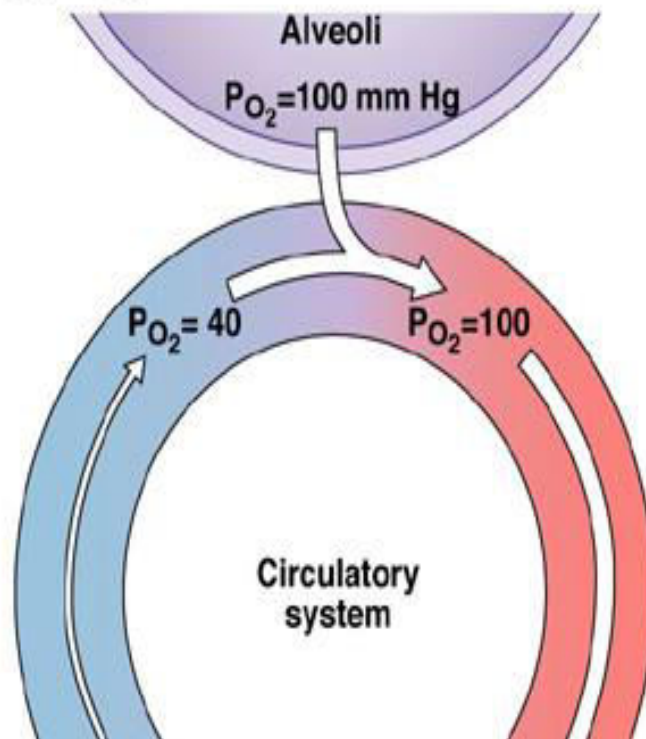
Objectives

- ⊙ Composition of air in different parts of respiratory system and difference in the composition of atmospheric air and alveolar air
- ⊙ Factors affecting diffusion of gases through respiratory membrane
- ⊙ Diffusion limited and perfusion limited gas exchange
- ⊙ Definition of diffusion capacity, the difference between the diffusion capacity of O₂ and that of CO₂ in the lungs.
- ⊙ Effect of V/Q on alveolar gas concentration

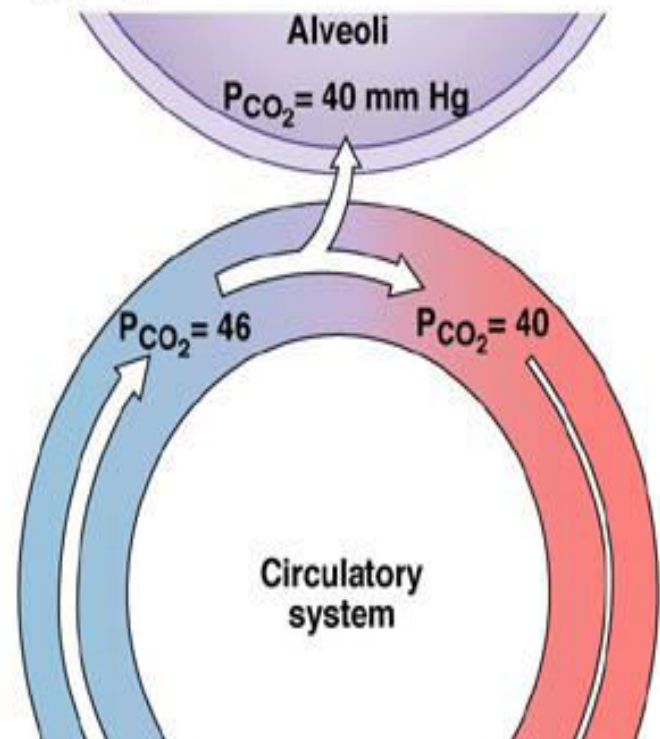
Gas exchange

- ⦿ Is a continuous process
- ⦿ Occurs through alveolo-capillary membrane by simple diffusion due to differences in partial pressure

(a) Oxygen diffusion



(b) CO₂ diffusion



Composition of air in different parts of respiratory system

Pressure mm Hg	Dry atmosph. air	Inspired air	Dead space air	Alveolar air
PO ₂	160	$0.21 \times 754 = 158$	$0.21 \times 713 = 149.7$	100
PCO ₂	0.3	$0.004 \times 754 = 0.3$	$0.004 \times 713 = 0.3$	40
PH ₂ O		5.7	47	47
PN ₂	600	$0.79 \times 754 = 596$	$0.79 \times 713 = 563$	573
Total	760	760	760	760

Reasons for the Difference in the Composition of Atmospheric Air and Alveolar Air

- ⊙ Constant absorption of O₂ from alveoli to pulmonary capillaries & diffusion of CO₂ from pulmonary capillaries to alveoli.
- ⊙ Humidification of atmospheric air as it passes through the respiratory passages leads to dilution of gases.
- ⊙ Partial replacement of alveolar air by atmospheric air with each breath.

Mechanisms by which Composition of Alveolar Air is Kept Constant ($P_{O_2} = 100\text{mHg}$, $P_{CO_2} = 40\text{mHg}$)

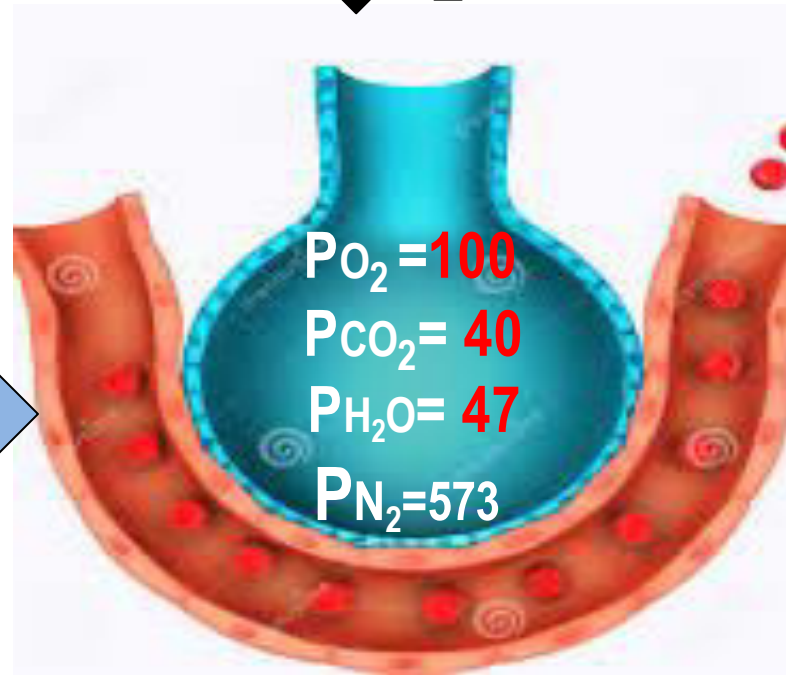
- 1) By proper ventilation and perfusion of lung:
 - O_2 continuously diffuses out of the alveoli into the blood stream and CO_2 continuously diffuses into the alveoli from blood
 - Inspired air mixes with alveolar air, replacing the O_2 and diluting the CO_2 .
- 2) Because of FRC of about 2 L at the end of expiration, 350 mL of inspired air or expired air has little effect on PO_2 and PCO_2 of alveolar air and alveolar gas composition remains constant.
- 3) Central and peripheral control mechanisms also operate to maintain alveolar gas composition constant.

Inspired air

$P_{O_2} = 158$
 $P_{CO_2} = 0.3$
 $P_{H_2O} = 5.7$
 $P_{N_2} = 596$

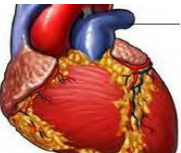
Expired air

$P_{O_2} = 116$
 $P_{CO_2} = 32$
 $P_{H_2O} = 47$
 $P_{N_2} = 565$



Pulmonary a

Pulmonary v



$P_{O_2} = 97$

Veins

$P_{O_2} = 40$
 $P_{CO_2} = 46$
 $P_{H_2O} = 47$
 $P_{N_2} = 573$

Tissue

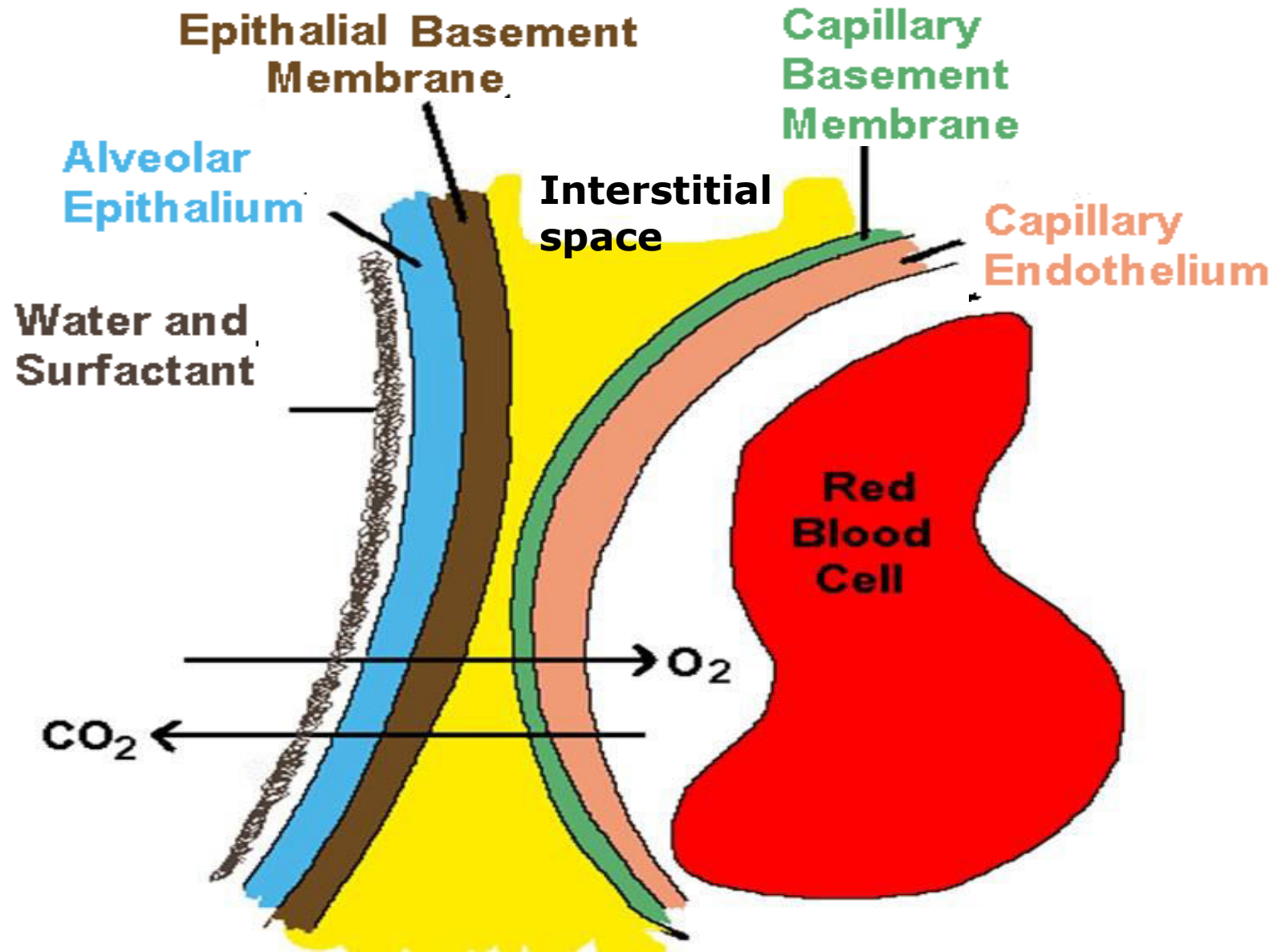
$P_{O_2} = <40$
 $P_{CO_2} = >46$
 $P_{H_2O} = 47$
 $P_{N_2} = 573$

Arteries

$P_{O_2} = 95$
 $P_{CO_2} = 40$
 $P_{H_2O} = 47$
 $P_{N_2} = 573$

Diffusion of gases through Respiratory Membrane

Layers of the respiratory membrane



Factors that affect rate of gas diffusion through the respiratory membrane

1) Thickness of respiratory membrane

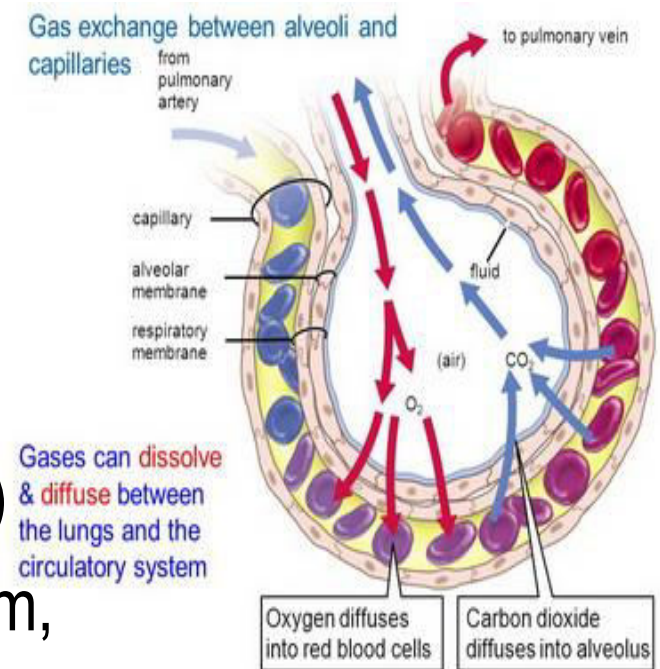
⊙ 2 Factors increases the rate of diffusion through the respiratory membrane

1) Thin respiratory membrane ($0.6 \mu\text{m}$)

- \uparrow Thickness \Rightarrow \downarrow rate of diffusion (e.g. pulmonary edema & fibrosis)

2) Diameter of pulmonary capillary = $8 \mu\text{m}$, diameter of RBC = $7.2 \mu\text{m}$

- RBC are squeezed through pulmonary capillary \rightarrow in close contact with respiratory membrane



2) Surface area of respiratory membrane:

- ⊙ Rate of diffusion is directly proportional to the surface area of respiratory membrane (70 m²)
 - ↓ Emphysema, chronic smokers and pneumectomy

3) Partial pressure difference of gases

- ⊙ ↑ Partial pressure gradient → ↑ rate of diffusion
 - Gases diffuse from a region of higher partial pr to a region of lower partial pr across the membrane until the pr of the gases on the two sides become equal

	Alveolar air	Pulmonary capillary blood	Partial pressure gradient
PO ₂ mm Hg	100	40	60
PCO ₂ mm Hg	40	46	6

4) Diffusion coefficient

- ⊙ Definition: Volume of gas in ml which diffuses through 1cm² of a membrane in one minute when there is a pressure difference of 1mm Hg across the membrane

$$\text{Diffusion coefficient} \propto \frac{S}{\sqrt{M.Wt}}$$

- ⊙ ↑Diffusion coefficient ⇒ ↑ rate of gas diffusion
- ⊙ Diffusion coefficient of O₂=1, CO₂=20, N₂=0.5, CO=0.8, He=0.9 (The diffusion coefficient of CO₂ is 20 times more than that of O₂)

$$D \propto \frac{\Delta P \times A \times S}{d \times \sqrt{M.Wt}}$$

- ⊙ D=Rate of diffusion of the gas, ΔP= Pressure gradient, A=Surface area, S=Solubility of the gas, d=Thickness of the respiratory membrane, MW= Molecular weight of the gas.

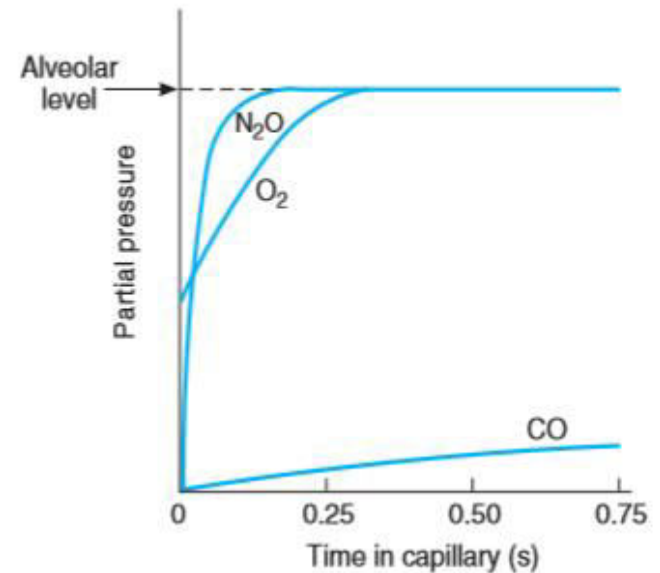
Perfusion- limited & diffusion limited gas exchange

- ⊙ Depends on their reaction with substances in the blood
- ⊙ Blood takes 0.75 sec to traverse the pul capillaries at rest

Perfusion limited exchange

⊙ Aesthetic gas nitrous oxide (N_2O)

- Doesn't form bond with Hb
- $\uparrow N_2O$ content of blood \Rightarrow rapid $\uparrow P_{N_2O}$ (equilibrium within 0.1 sec)
- Diffusion of N_2O can be increased only if perfusion increases



⊙ Oxygen (O_2)

- Reach equilibrium with within 0.3 sec (perfusion limited)
- In fibrosis (thickening of resp membrane) & emphysema (\downarrow surface area of resp membrane) \Rightarrow $\downarrow O_2$ diffusion (diffusion limited)

Diffusion limited exchange

⊙ Carbon monoxide (CO)

- Strong bond to Hb \rightarrow \uparrow CO in blood \Rightarrow minimum \uparrow P_{co}
- Equilibrium is not reached in 0.75 sec
- Transfer of CO is limited by the rate of diffusion, not the amount of blood available

Perfusion limited gases	Diffusion limited gases
N ₂ O (anesthetic gas)	CO
CO ₂	
O ₂ (Normal condition)	O ₂ (Emphysema, fibrosis, exercise)

The diffusion capacity of the lung to the gases:

⊙ **Definition:** The volume of gas which is diffused/min/1 mmHg difference in partial pressure of the gas.

⊙ **Measurement:** The diffusion capacity for CO (D_{LCO}) is measured as an index of diffusion capacity because its uptake is diffusion limited.

- D_{LCO} is proportional to the amount of CO entering the blood (V_{CO}) divided by P_{CO} in the alveoli (P_{ACO}) minus the partial pressure of CO in the blood entering pulmonary capillaries \approx zero (except in habitual smokers)

$$D_{LCO} = \frac{V_{CO}}{P_{ACO} - P_{aCO}} \rightarrow D_{LCO} = \frac{V_{CO}}{P_{ACO}}$$

⊙ **Factors:** same factors that affect rate of gas diffusion through the respiratory membrane affects diffusion capacity of the lung

- It is directly proportional to the surface area of alveolo-capillary membrane and inversely proportional to its thickness.

⊙ Normal value

⊙ D_{LCO} at rest is 25mL/min/mmHg

- It increases to three fold during exercise because of capillary dilation and an increase in the number of active capillaries

⊙ $D_{LO_2} = D_{LCO} = 25\text{mL/min/mm Hg}$

- $\uparrow D_{LO_2}$ in Exercise
- $\downarrow D_{LO_2}$ Diseases (fibrosis of alveolar walls)

⊙ $D_{LCO_2} = 400\text{ml/min/mm Hg} (> D_{LO_2})$

- High solubility of CO_2 in cell membrane (CO_2 retention is rarely a problem in patients with alveolar fibrosis even when the reduction in diffusion capacity for O_2 is sever)

Effect of V/Q on alveolar gas concentration

- ⊙ Ratio of alveolar ventilation(V) to pulmonary blood flow (Q)
 - Matching ventilation and perfusion is important to achieve the ideal exchange of O₂ and CO₂
 - Normal V/Q (whole lung) at rest is 0.8 (4L/min ÷ 5L/min)

Ventilation	Normal	Zero	Normal
Perfusion	Normal	Normal	Zero
V/Q	Normal	Zero	Infinity
Situation	Normal	Complete airway obstruction →shunted blood	Pulmonary artery obstruction → dead space
Gas exchange	Optimal	No gas exchange	No gas exchange
Alveolar:Po ₂ mmHg Pco ₂ mmHg	Po ₂ = 100 Pco ₂ = 40	Po ₂ = 40 Pco ₂ = 46	Po ₂ = 149.7 Pco ₂ = 0.3